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Diisooctyl phthalate, the major secondary metabolite of *Bacillus subtilis*, could be a potent antifungal agent against *Rhizoctonia solani*: GC-MS and *in silico* molecular docking investigations



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Abstract

Bioactive secondary metabolite compounds produced by microorganisms, including bacteria, have certain functions, one of which is to act as antifungal agents. The current study examined the ability of seventeen *Bacillus* isolates to suppress the soilborne fungus *Rhizoctonia solani*, determining their potential as biocontrol agents. The KSAR2 isolate was the most effective at inhibiting *R. solani* mycelial growth, with a 61.2% inhibition rate achieved using the dual culture method. The analysis of the nucleotide sequencing of the 16S rRNA gene indicated that the isolate KSAR2 was identified as *Bacillus subtilis* and has been deposited in GenBank with the accession number PQ238901. The gas chromatography-mass spectrometry investigation revealed that the most abundant secondary metabolite compounds of KSAR2 were diisooctyl phthalate (38.18%), dibutyl phthalate (9.41%), tris(2,4-di-tert-butylphenyl) phosphate (7.19%), dotriacontane (6.67%), docosane (6.62%), and hexadecanoic acid (4.77%). A study of molecular docking interactions also found that diisooctyl phthalate had the highest binding energy with the fungal chitin synthase, measuring -7.90 kcal/mol. This indicates that interaction with this particular protein could elucidate the mechanism underlying the assessed antifungal activity. Finally, the present study showed that the rhizobacterium *B. subtilis* strain KSAR2 can inhibit the growth of *R. solani*. This makes it a promising biocontrol agent for protecting plants against fungal infestations. Furthermore, diisooctyl phthalate may function as a natural, bioactive fungicide by inhibiting chitin synthase in *R. solani* and potentially other fungi. This could consequently lead to important advancements in the development of novel and potent antifungal agents.

Keywords: GC-MS; molecular docking; diisooctyl phthalate; chitin synthase; Rhizoctonia solani; Bacillus subtilis.

1. Introduction

The secondary metabolites generated by various living microorganisms display an impressive diversity in their chemical structures and biological activities [1]. These secondary metabolites are made up of different compounds with low molecular weights that are produced during secondary metabolism. These include alkaloids, terpenoids, peptides, and polyketides, and they happen separately from the primary microbial growth phase. Each compound exhibits a range of biological activities. Several have already been recognized for their roles as antivirals, antimicrobials, antioxidants, and different biological activities. Some are utilized as colorants, pesticides, or growth enhancers for animals or plants [2–4]. A common soilborne necrotroph,

Rhizoctonia solani, causes harm to numerous commercially significant crops. Members of this species differ greatly in terms of colony appearance, biochemical and molecular markers, pathogenicity, and aggressiveness. This diversity has allowed for the classification of the species into 14 somatically incompatible groups, also known as anastomosis groups. *R. solani* can infect a variety of hosts and cause several symptoms, like damping-off, seedling blight, stem canker, black scurf, hypocotyl rot, stem rot, crown rot, and limb rot [5]. The diseases produced by *R. solani*, which cause damping-off, wilt, and root rot of several crops, were managed using a variety of approaches, including breeding programs, agricultural methods, broad-spectrum

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fungicides, and biological control. Using chemical fungicides may be more effective than other methods in controlling diseases caused by *R. solani*, but it also creates an unevenness in the microbial community that is detrimental to the activity of beneficial organisms, pollutes the environment, and harms mammals and other beneficial living things [6]. Therefore, it is essential to conduct comparative studies between bioagents and conventional fungicides to identify the most effective microbe that can reduce the severity of plant diseases and assess its potential [7].

Biocontrol is regarded as one of the most efficient, secure, and beneficial methods for the environment [8,9]. Due to its high efficacy and environmental friendliness, the use of biocontrol agents is gaining attractiveness as an alternative approach for managing plant diseases [10]. Rhizobacteria that promote plant growth (PGPR) are well known for being the most efficacious biocontrol agents for controlling plant diseases. The Bacillus genus includes many endospore-forming, rod-shaped, and catalase-positive bacteria that can be isolated from a wide range of biological environments, such as soil, compost, the rhizosphere, fermented food, and clinical specimens [11]. Biocontrol of plant diseases has relied on Bacillus species and their related compounds for a long time because they release a wide range of helpful chemicals, can colonize roots better, and can make spores [12]. It has been found that many Bacillus species exhibit potent inhibitory activity against plant diseases and stimulate plant growth; as a result, over eighteen commercial products derived from Bacillus species have been sold all over the world [11].

Gas chromatography coupled with mass spectrometry (GC-MS) serves as a crucial instrument for finding new natural products because it can identify and analyze different chemical parts in microbial samples, such as organic acids, amino acids, alkaloids, and flavonoids [13-15]. Using GC-MS to look at extracts from different types of Bacillus makes it easier to find a wide range of metabolites, which can then be tested to see what biological effects they have [16]. This analytical approach provides essential insights into the metabolite composition of Bacillus spp., laying the groundwork for understanding their potential biological properties [17]. In silico predictions of pharmacological, pharmacokinetic, and toxicological outcomes are made using computational prediction models. These models significantly influence the selection of methods that propel advancements in both pharmaceuticals and technology [18]. Molecular docking is a cost-effective and efficient method for the development and evaluation of medicines. This method gives information about how drugs and receptors work together, which can be used to guess

how the drug model will bind to target proteins, ensuring stable binding at the ligand binding sites [19,20]. Recently, it was utilized in the field of plant pathology to forecast the potential bioactive compounds that could combat plant pathogens [21].

In this context, the current study aimed to identify which of the isolated Bacillus isolates could serve as biocontrol agents against R. solani. The promising Bacillus isolate was further characterized at the molecular level by 16S rRNA analysis. Furthermore, bioactive secondary metabolite compounds present in bacterial culture filtrate were identified by GC-MS analysis. Subsequently, a computational molecular docking analysis was conducted to pinpoint bioactive compounds that may exhibit antifungal properties. This study seeks to predict the structures and formulas of bioactive compounds, which could be critical for the discovery and development of new antifungal formulations.

2. Materials and methods

2.1. Isolation experiments

In the Kingdom of Saudi Arabia (KSA), soil dilution from tomato, potato, pepper, and cucumber rhizospheres was used for isolation experiments. For fourteen minutes, one gram of soil was shaken in nine milliliters of sterile saline (0.85% NaCl) solution. 200 μ L of the 10⁻⁵, 10⁻⁶, and 10⁻⁷ dilutions were applied to the glycerol nutrient agar media and the peptone nutrient agar medium [22]. After 48 hours of incubation at 30°C, many colonies of bacteria were visible. The observed colonies were separated and purified using the single colony isolation procedure. Bacterial cultures were kept in 50% glycerol broth at -80°C for further studies [23].

2.2. Antagonism test of *Bacillus* isolates

The antagonistic impact of Bacillus isolates as biocontrol agents against R. solani, a plant pathogenic fungus, was examined using an in vitro antagonist test. A streak line of the antagonistic bacterial isolates, cultured for two days, was inoculated onto potato dextrose agar (PDA) plates. This was done 48 hours before any testing fungi were inoculated. In the center of the Petri plate was a mycelial circle (5 mm in length) holding an energetically growing culture of the fungus under investigation, always at a constant space from the other edge, and incubated at 30°C for three to seven days. The method for calculating the percentage of inhibition was as outlined using Maurhofer et al. [24]. The formula growth inhibition rate (%) = $[(R-r)/R] \times$ 100 was used to calculate the percentage of inhibition in mycelium growth. In this formula, r represents the fungal development radius towards the bacterial treatment, while the value R represents the fungal colony's radius measurement on the control plate.

2.3. Molecular characterization of promising *Bacillus* isolate

The pure bacterial colony was injected into 5 mL of Luria-Bertani broth and incubated at 30 °C overnight with agitation. Following the manufacturer's guidelines, the Wizard Genomic DNA Purification Kit (Promega, USA) was employed to extract bacterial genomic DNA from a 1 mL bacterial pellet. The integrity and quality of the extracted DNA were assessed via 1.2% agarose gel electrophoresis, while (A260/A280) and concentration were purity measured using the SPECTRO Star Nano apparatus (BMG Labtech, Germany). The extracted genomic DNA was corrected to 50 ng/µL before being used as a template in PCR. To amplify the 16S rRNA gene, we used universal primers 27F (5'-AGAGTTTGATCCTGGCTCAG-3') and 1492R (5'-CGGCTACCTTGTTACGACTT-3'). The 20-µL PCR reaction mixture had 10 µL of PCR master mix, 1 µL of DNA, 1 µL of each primer at a concentration of 10 pmol/µL, and 7 µL of double-sterile distilled water. The PCR protocol started with an initial denaturation at 95°C for 4 minutes, followed by 35 cycles of denaturation at 94°C for 45 seconds, annealing at 50°C for 45 seconds, and extension at 72°C for 90 seconds. The last extension phase was carried out at 72°C for 5 minutes. Next, 5 µL of PCR products were separated using 2% agarose gel electrophoresis and photographed with a gel documentation system. The PCR product was sequenced immediately after purification using the QIAquick gel extraction kit (Qiagen Inc.). Sanger sequencing of PCR products was carried out using the BigDye® Terminator v3.1 Cycle Sequencing kit (Applied Biosystems) and the model 3130xl Genetic Analyzer (Applied Biosystems). The collected DNA nucleotide sequences were validated using NCBI-BLAST (http://blast.ncbi.nlm.nih.gov/Blast.cgi). The annotated nucleotide sequence was analyzed and submitted to GenBank to obtain an accession number.

2.4. Gas chromatograph-mass spectrometer analysis of secondary metabolites

The GC-MS approach was utilized to identify the active biomolecules in the culture filtrate of isolate BR2. The culture filtrate was obtained by centrifuging the culture generated during the batch fermentation process. The culture filtrate was mixed with ethyl acetate in equal volumes (1:1 ratio) and stirred vigorously for 20 minutes. The ethyl acetate phase was isolated and concentrated with a rotary evaporator set at a temperature of 50 °C. The TRACE 1300 Series GC-MS (Thermo, USA) equipped with a split mode mass detector was utilized to identify secondary metabolite compounds. The transport was helium gas, and the flow rate was 1 mL/min. The column oven was initially maintained at a temperature of 45 °C. Then it was programmed to increase by 5 °C every minute until it reached 200 °C, and then it would stay there for 5 minutes. From there, it increased by 5 °C per minute until it reached 300 °C. The injector and MS transfer line maintained constant temperatures of 270 and 250 °C, respectively. Setting up a 3-minute solvent delay on the autosampler AS1310 in split mode, 1 µL portions of diluted samples were automatically fed into the GC machine. The electricity of 70 eV and an ion source at 200 °C were used to get electron impact (EI) mass spectra in full scan mode. The spectra ranged from 40 to 500 m/z. Parts were located by comparing mass spectra and retention durations to those in the WILEY 09 and NIST 11 databases [4].

2.5. Molecular docking assessment

The research collaboratory for structural bioinformatics protein data bank (RCSB PDB; https://www.rcsb.org/), AlphaFold (https://alphafold.ebi.ac.uk/), and the Molecular Operating Environment (MOE 2022.02, Chemical Computing Group, Montreal, QC, Canada) tools were utilized for the retrieval of proteins and ligands, as well as for molecular docking. The three-dimensional (3D) structures of secondary metabolite molecules (ligands) were obtained from the PubChem database (https://pubchem.ncbi.nlm.nih.gov/) in structured data file (SDF) format and subsequently analyzed using molecular operating environment (MOE) software. The three-dimensional structures of fungal chitin synthase were obtained from the universal protein resource (UniProt) database (https://www.uniprot.org/). The target protein was prepped for docking using UCSF Chimera software by eliminating water and ligand molecules from the protein structures, in addition to performing energy minimization on the target protein. The target protein was docked with ligands by determining the binding site, employing the induced fit model for docking, and visualizing the results using the same program [25].

2.6. Statistical analysis

The results are presented as the average and the standard deviation (average \pm SD), with the experiments being performed three times. Tukey's honest significant differences (H.S.D.) test was used to look for differences and find statistical significance. A significance level of $p \le 0.05$ was set after the one-way analysis of variance (ANOVA) of the data using CoStat software (version 6.45; https://cohortsoftware.com/costat.html; 2022). We arranged the letters in descending order, with the relationship (a > b > c) indicating statistical significance. appeared The letters to be indistinguishable from one another.

3. Results

3.1. Fungal and bacterial isolates

The pathogenic *R. solani* strain R11 (Accession number OP456528) previously isolated was used in this study [26]. In the Kingdom of Saudi Arabia

(KSA), an isolation experiment was performed to isolate *Bacillus* isolates from the rhizosphere of tomato, potato, pepper, and cucumber. In the present investigation, seventeen *Bacillus* isolates were isolated and preserved. BR1, BR2, BR4, BR5, and BR6 *Bacillus* isolates were initially part of group 1, which was isolated from the tomato rhizosphere. Second, BR8, BR9, BR12, BR13, and BR14 were part of group 2, which was separated from the potato rhizosphere. Thirdly, BR15, BR16, BR17, and BR18 were extracted from the pepper rhizosphere by group 3. Finally, BR19, BR20, and BR21 were included in group 4, which was isolated from a cucumber rhizosphere.

3.2. Antagonism test of Bacillus isolates

The antagonistic influence of seventeen *Bacillus* isolates on the *in vitro* development of *R. solani* was

investigated as a possible bioagent. There were notable variations between the Bacillus isolates, according to the findings from the current investigation (Figure 1). With the highest percentage of inhibition against the pathogen, 61.2%, isolate BR2 was shown to be more efficient than other isolates in stopping the fungal mycelia growth of R. solani. The next highest percentage of inhibition against the pathogen was reported by bacterial isolate BR6, which was 57.6%. The percentages of inhibition for the Bacillus isolates BR1, BR4, and BR9 were close to each other, which recorded 49%, 52.9%, and 50.2%, respectively. Isolate BR18 exhibited the lowest percentage of inhibition (5.9%) against R. solani. However, there were no antagonistic behaviors displayed by isolates BR8, BR14, BR19, BR20, and BR21 (Figure 2).



Fig. 1. Effect of seventeen *Bacillus* isolates on the mycelial development of *R. solani* utilizing the dual culture method. Each column displays the mean value of three duplicated measurements, while the bars represent the standard deviation. Tukey's HSD test, utilizing a significance threshold of 0.05, indicates that values within each column designated by the same letter (a/b/c/d/e) demonstrate no statistically significant differences.



Fig. 2. Antifungal activities of seventeen *Bacillus* isolates against *R. solani* growth. Each column displays the mean value of three duplicated measurements of growth inhibition percentage, while the bars represent the standard deviation. Tukey's HSD test, utilizing a significance threshold of 0.05, indicates that values within each column designated by the same letter (a/b/c/d/e) demonstrate no statistically significant differences.

3.3. Molecular characterization of *Bacillus* isolate BR2

The morphological characteristics of Bacillus isolate BR2 were gram-positive and endospore-forming. On nutrient agar plates, colonies appeared creamy-white and rough, and their edges were slightly irregular. Bacillus isolate BR2 was molecularly characterized by sequencing the 16S rRNA gene. The PCR amplification yielded a product of approximately 1500 base pairs. Based on NCBI-BLAST alignment and phylogenetic tree analysis, the BR2 isolate has been determined to be B. subtilis. The annotated nucleotide sequence of 1448 bp has been submitted in the GenBank database under the name B. subtilis strain KSAR2, with the accession number PQ238901. The nucleotide sequence showed 100% sequence coverage with other Bacillus species that were deposited in GenBank. The KSAR2 was closely related to other B. subtilis strains, especially China isolate (Acc# OR335070) by 98.76% identity.

3.4. Identification of bioactive metabolites of *Bacillus subtilis*

Extract of KSAR2 culture filtrate, which was achieved by ethyl acetate, was analyzed using GC-MS analysis (Figure 3) to identify the specific compounds that could be responsible for its antagonistic effect against R. solani. The GC-MS tests showed that the KSAR2 culture filtrate contained 22 different compounds (Table 1). These compounds, including diisooctyl phthalate (38.18%), dibutyl phthalate (9.41%), tris(2,4-di-tertphosphate (7.19%), dotriacontane butylphenyl) (6.67%), docosane (6.62%), hexadecanoic acid (4.77%),7,9-ditertbutyl-1-oxaspiro(4,5)deca-6,9diene-2,8-dione (3.06%), 2,5-Piperazinedione,3,6-RT: 0.00 - 45.28 SM: 15B

bis(2-methylpropyl) (2.04%), hexadecanoic acid,2hydroxy-1-(hydroxymethyl) ethyl ester (2.04%), methanone,(1-hydroxycyclohexyl)phenyl (1.9%), octadecanoic acid (1.63%), 3,4-dihydro-2h-1,5-(3"-tbutyl) benzodioxepine (1.42%), hexadecane,2,6,10,14-tetramethyl (1.32%), octadecanoic acid, 2-hydroxyethyl ester (1.29%), and heptacosane (0.82%).

3.5. Molecular docking interaction investigation

The molecular docking analysis was performed to determine the potential binding interactions of the identified secondary metabolite compounds from B. subtilis strain KSAR2 with the chitin synthase of R. solani as the target protein. A total of sixteen metabolite compounds were successfully docked into the chitin synthase, with docking scores ranging from -4.20 to -7.90 kcal/mol (Table 2). Among them, diisooctyl phthalate and estra-1,3,5(10)-trien-17á-ol demonstrated a significant affinity for the binding site of chitin synthase, with docking scores of -7.90 and -7.20 kcal/mol, respectively (Figure 4). On the other hand, hexadecanoic acid and hexadecane showed the lowest binding affinity with the same docking score of -4.20 kcal/mol (Table 2). Figure 4 presents the three-dimensional visualizations of ligand-protein complex interactions between the two compounds and chitin synthase. The binding energy of diisooctyl phthalate predominantly arises from hydrophobic interactions with active site residue amino acids, including alkyl interactions with LEU136, LEU347, LEU852, LEU351. ILE848, LEU853. and Furthermore, estra-1,3,5(10)-trien-17á-ol exhibited interactions with VAL392 and ALA440 via hydrophobic interactions.



Fig. 3. GC-MS chromatography of ethyl acetate extract of the culture filtrate of Bacillus subtilis strain KSAR2.

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| , | Table 1 | . The list of detected compounds i | n the ethyl ac | etate extract of | f culture filtrate of B. subtilis strain KS | AR2. |
|---------|-------------|--------------------------------------------------|------------------------------------------------|---------------------|--------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|----------------|
| RT | Area | Compound Name | Molecular | Molecular Weight | Compound structure | CAS |
| 35.82 | 38.18 | Diisooctyl phthalate | C ₂₄ H ₃₈ O ₄ | 390 | -< | 131-20-4 |
| | | | | | | |
| | | | | | a for the second | |
| | | | | | | |
| 25.63 | 9.41 | Dibutyl phthalate | $C_{16}H_{22}O_4$ | 278 | | 84-74-2 |
| | | | | | | |
| 12.00 | 7 10 | | C H O D | | | 0500 (1 |
| 42.90 | 7.19 | Tris(2,4-ditert-butylphenyl)phosphate | $C_{42}H_{63}O_4P$ | 662 | | 95906-1 1-9 |
| | | | | | | |
| | | | | | | |
| | | | | | XXX X | |
| 41.84 | 6.67 | Dotriacontane | $C_{32}H_{66}$ | 450 | | 544-85-4 |
| | | | | | | |
| 33.54 | 6.62 | Docosane | $C_{22}H_{46}$ | 310 | | 629-97-0 |
| | 1.55 | | | | | 55.10.0 |
| 26.44 | 4.77 | Hexadecanoic acid | $C_{16}H_{32}O_2$ | 256 | | 57-10-3 |
| | | | | | | |
| 24.75 | 3.06 | 7,9-di-tert-butyl-1-oxaspiro (4,5)deca- | C17H24O3 | 276 | • | 82304-66-3 |
| | | 6,9-diene-2,8-dione | | | | |
| | | | | | | |
| 25.10 | 2.04 | | C U NO | 226 | | 1426 07 7 |
| 25.10 | 2.04 | 2,5-Piperazinedione,3,6-bis(2- methylpropyl)- | $C_{12}H_{22}N_2O_2$ | 226 | NH | 1436-27-7 |
| | | | | | | |
| 25.07 | 2.04 | | C U O | 220 | | 542.44.0 |
| 35.27 | 2.04 | dihydroxypropyl ester | $C_{19}H_{38}O_4$ | 330 | | 542-44-9 |
| | | | | | | |
| 10.00 | 1.00 | | C U O | 204 | | 047 10 2 |
| 19.99 | 1.90 | Methanone, (1- hydroxycyclohexyl)phenyl- | $C_{13}H_{16}O_2$ | 204 | CH CH | 947-19-3 |
| | | | | | | |
| 30.03 | 1.63 | Octadecanoic acid | C ₁₈ H ₃₆ O ₂ | 284 | u0 | 57-11-4 |
| | | | | | | |
| | | | | | | |
| 16.62 | 1.42 | 3,4-dihydro-2h-1,5-(3"-t-butyl) | $C_{13}H_{18}O_2$ | 206 | | NA |
| | | benzouroxepine | | | | |
| | | | | | O^ | |
| 21.62 | 1.32 | Hexadecane,2,6,10,14-tetramethyl- | $C_{20}H_{42}$ | 282 | | 638-36-8 |
| | | | | | | |
| 38 36 | 1 29 | Octadecanoic acid 2-hydroxyethyl | $C_{20}H_{40}O_2$ | 328 | | 111-60-4 |
| 50.50 | 1.2) | ester | 020114003 | 320 | √√√√√√√ ун | 111 00 1 |
| | | | | | | |
| 36.86 | 0.82 | Heptacosane | C27H56 | 380 | | 593-49-7 |
| | | | -2750 | | | |
| 31.02 | 0.75 | Hexadecanoic acid, 2- | $C_{18}H_{36}O_3$ | 300 | | 4219-49-2 |
| | | nydroxyetnylester | | | nu v v v v v v v | |
| 25.40 | 0.56 | 6 other 5 hydroxy 2 2 7 | СЦО | 202 | Он о | NA |
| 23.40 | 0.30 | trimethoxynaphthoquinone | $C_{15}\Pi_{16}O_{6}$ | 292 | | INA |
| <u></u> | 0.15 | | <u> </u> | 1.5.0 | · · · · · · · · · · · · · · · · · · · | |
| 24.49 | 0.48 | Docosane, 11-decyl- | $C_{32}H_{66}$ | 450 | | 55401-55-3 |
| | | | | | | |
| | | | | | 5 | |

DIISOOCTYL PHTHALATE, THE MAJOR SECONDARY METABOLITE OF BACILLUS SUBTILIS, 1143

| 23.61 | 0.43 | 2,5-di-tert-butyl-1,4-benzoquinone | $C_{14}H_{20}O_2$ | 220 | 2460-77-7 |
|-------|------|------------------------------------|-----------------------------------|-----|------------|
| 27.70 | 0.40 | Estra-1,3,5(10)-trien-17á-ol | C ₁₈ H ₂₄ O | 256 | 2529-64-8 |
| 17.82 | 0.38 | Hexadecane | C ₁₆ H ₃₄ | 226 | 544-76-3 |
| 21.09 | 0.33 | Tetradecane, 2,6,10-trimethyl- | $C_{17}H_{36}$ | 240 | 14905-56-7 |

Table 2. Molecular docking score of secondary metabolites compounds of *B. subtilis* strain KSAR2 with chitin synthase of *R. solani*.

| synthase of R. soluhi. | | | | |
|-----------------------------------------|--------------------------|--|--|--|
| Compound Name | Docking score (kcal/mol) | | | |
| Diisooctyl phthalate | -7.90 | | | |
| Estra-1,3,5(10)-trien-17á-ol | -7.20 | | | |
| Tris(2,4-ditert-butylphenyl)phosphate | -6.50 | | | |
| Docosane, 11-decyl- | -6.30 | | | |
| Dibutyl phthalate | -6.20 | | | |
| Heptacosane | -6.20 | | | |
| 2,5-di-tert-butyl-1,4-benzoquinone | -6.10 | | | |
| Octadecanoic acid | -5.60 | | | |
| Octadecanoic acid, 2-hydroxyethyl ester | -5.50 | | | |
| Methanone,(1-hydroxycyclohexyl)phenyl- | -5.40 | | | |
| Docosane | -5.20 | | | |
| Dotriacontane | -5.10 | | | |
| Hexadecane, 2, 6, 10, 14-tetramethyl- | -4.50 | | | |
| Tetradecane, 2,6,10-trimethyl- | -4.50 | | | |
| Hexadecanoic acid | -4.20 | | | |
| Hexadecane | -4.20 | | | |



Diisooctyl phthalate; Binding free energy = -7.90 kcal/mol

Estra-1,3,5(10)-trien-17á-ol; Binding free energy = -7.20 kcal/mol

Fig. 4. The three-dimensional of molecular docking interactions of diisooctyl phthalate (left) and estra-1,3,5(10)-trien-17á-ol (right) with chitin synthase of *R. solani*.

4. Discussion

R. solani is a soil-borne fungus capable of producing sclerotia, allowing it to survive longer in the soil. It has a broad host range. As a result, managing *R. solani* is somewhat challenging. The majority of the pesticides employed to manage *R. solani* have harmful effects on non-targeted organisms as well as

the environment, making them ineffective [27]. Because of growing consumer concerns about chemical pesticides remaining in food and environmental safety, there is a need to explore alternative approaches, such as biocontrol [26,28]. According to Huang et al. [29], *Bacillus* strains are crucial biocontrol agents against diseases caused by *R. solani* on pepper, potato, tomato, and other plants.

Bacillus strains possess not only the capability to impede the growth of fungal pathogens via contactdependent antimicrobial metabolites but also the capability to submerge hydrolases such as protease, glucanase, chitinase, xylanase, and cellulase. These hydrolases target the glycosidic bonds of the core constituents of the fungal pathogen's cell wall, which include protein, glucan, and chitin. Additionally, they can secrete volatile organic compounds (VOCs), or antimicrobial metabolites, such as benzaldehyde, 1,2benzothiazole-3(2H)-one, 1,3-butadiene, 2-nonanone, 2-decanone, and 2-propanone, which can distance inhibit target infections [30]. In this study, we assessed seventeen isolates of Bacillus as possible biocontrol agents against R. solani's in vitro mycelial growth. With an inhibition percentage of 61.2%, KSAR2 outperformed the other Bacillus isolates in its inhibitory effectiveness against R. solani mycelia. The antagonistic impact of the bioagent, KSAR2 may be referred toward its excretion of enzymes and chemicals that have direct and indirect antifungal properties. Matar et al. [31] tested fourteen B. subtilis isolates for their ability to inhibit R. solani isolated from cotton. They showed four B. subtilis isolates had a more antagonistic effect against R. solani. B. subtilis strain B4 gave a greater inhibition percentage equal to 47.1% compared to the other strains against R. solani of peanut [32]. As stated by Huang et al. [29], B. subtilis effectively suppressed the mycelial development of R. solani, with a 42.3% inhibition percentage. Chen et al. [33] revealed that the antifungal substances in the filtrate of B. subtilis culture could cause more protoplasm leaks, cell-wall fragmentation, and cytoplasmic vacuoles in the R. solani mycelium.

Bioactive compound profiling revealed that there were 22 active compounds detected in B. subtilis strain KSAR2. However, there were six major bioactive compounds with the highest percentage of peak area. The most important compound found was diisooctyl phthalate. It was followed by dibutyl phthalate, tris(2,4-di-tert-butylphenyl) phosphate, dotriacontane, docosane, and hexadecanoic acid. Those bioactive compounds were previously described as antimicrobial and antifungal compounds. However, further purification and assays are required to confirm that those bioactive compounds were responsible for *B. subtilis* strain KSAR2 antifungal activity. Diisooctyl phthalate exhibits antibacterial and antifouling properties [34]. One of the most significant bioactive metabolites was formed by Paecilomyces sp. (AUMC 15510) [35] and Trichoderma harzianum TH7 [36]. It has properties that promote wound healing, as well as the ability to inhibit the growth of microorganisms [37,38]. Lykholat et al. [39] demonstrated that the diisooctyl phthalate, the main constituent of Penicillium sp.'s ethyl acetate extract, was responsible for its antifungal activity against Alternaria alternata. El-

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Enain et al. [34] have recently proposed that diisooctyl phthalate could potentially be used as a medicinal agent for the treatment of severe bacterial infections. Dibutyl phthalate had substantial inhibitory effects on the mycelial development of Botrytis cinerea [40]. Blazević et al. [41] demonstrated that dibutyl phthalate inhibited the development of gram-positive bacteria species Staphylococcus aureus and the yeast Candida. albicans. Furthermore, the antifungal activity of dibutyl phthalate, which is synthesized by B. amyloliquefaciens and B. velezensis, was shown [42]. Furthermore, Streptomyces has been found to exhibit antibacterial activity of dibutyl phthalate [43,44]. Dibutyl phthalate, a bioactive chemical synthesized by Streptomyces albidoflavus 321.2, exhibited significant antimicrobial action against both Gramnegative bacteria, and Gram-positive as well as unicellular and filamentous fungi. Dibutyl phthalate of Streptomyces strain BITDG-11 exhibits potent antifungal properties against Fusarium oxysporum f. sp. cubense [45]. Dibutyl phthalate has also been shown to be an antimicrobial bioactive compound in marine *Pseudomonas* strains [46,47] and other actinomycete strains [48,49]. Hexadecanoic acid is a fatty acid that has been found in many medicinal plants [50,51] as well as in *Bacillus* spp. metabolites [4,52]. It has been suggested that it may be antifungal. hypocholesterolemic, anticancer. antibacterial, antioxidant, nematicide, and pesticide [53]. The hexadecanoic acid synthesized by the demonstrated significant efficacy in Bacillus suppressing the growth of Colletotrichum gloesporoides [54]. Hexadecanoic acid can hinder the growth of fungi by creating intricate compounds by binding to the active components of fungal cell walls [55]. It has also been shown that the ethyl ester of octadecanoic acid, which was discovered in B. atrophaeus along with other chemicals, can effectively suppress Verticillium wilt [56]. Dotriacontane compounds exhibit antioxidant, antiplasmodic, and antimicrobial properties [57,58]. The essential oil extract contains 4.5% docosane and has been reported to have antibacterial properties [59]. Each of these compounds has previously been suggested to possess a specific antibacterial action [60]. Subsequent investigations will prioritize the examination of the precise antimicrobial mechanisms employed by these bioactive compounds.

The progress in computational science significantly influenced the identification of novel bioactive compounds [61]. Recently, the use of virtual screening methods in agriculture has become widespread and prevalent, significantly reducing the time and costs associated with the discovery and development of new antimicrobial agents [21,62]. Molecular docking is a way to find new ligands for protein structures. It is also a key part of using structure information to design antimicrobial agents

[63,64]. The development of novel antifungal agents is challenging due to the similarities in biological pathways between fungi and their hosts. Chitin synthases are enzymes associated with membranes that generate chitin, an essential structural element of fungal cell walls. It catalyzes chitin synthesis and transfer across the membrane, making it a target for antifungal agents. Blocking it leads to a targeted suppression of fungal growth or inhibition without adversely affecting vertebrates or plants [65,66]. In the context of plant infection, the degradation or prevention of pathogen chitin formation presents a promising approach to enhancing crop immunity against fungal infections [67]. In the current study, molecular docking interaction analyses were employed to investigate the binding capability of bioactive compounds from KSAR2 culture filtrate extract that could bind to fungal chitin synthase, which is a target for antifungal agents. A more negative docking score indicates a greater binding affinity between the receptor and ligand molecules, demonstrating the enhanced efficacy of bioactive substances [20]. This investigation identifies diisooctyl phthalate and estra-1,3,5(10)-trien-17á-ol as the primary compounds, exhibiting the highest docking scores of -7.90 and -7.20 kcal/mol, respectively, among the evaluated bioactive compounds. Interestingly, the docking analysis aligns with the GC-MS analysis, indicating that diisooctyl phthalate exhibited the highest peak area at 38.18%, along with the highest docking score of -7.90 kcal/mol. Therefore, diisooctyl phthalate may serve as bioactive compounds with the potential to act as antifungal agents against phytopathogenic fungi.

5. Conclusions

The current study demonstrated that the molecularly characterized B. subtilis strain KSAR2 (Accession number # PQ238901) exhibited notable antifungal activity against the soilborne fungus Rhizoctonia solani, achieving an inhibition rate of 61.2%. The GC-MS analysis of the ethyl acetate extract from the bacterial culture filtrate identified 22 unique compounds. The investigation detected a variety of secondary metabolite chemicals, with diisooctyl phthalate being the most common at 38.18%. Next, dibutyl phthalate was at 9.41%, tris(2,4-di-tertbutylphenyl) phosphate was at 7.19%, dotriacontane was at 6.67%, docosane was at 6.62%, and hexadecanoic acid was at 4.77%. The investigation of molecular docking interactions revealed that diisooctyl phthalate had the highest binding energy with fungal chitin synthase, quantified at -7.90 kcal/mol. Thus, the current study shows that the rhizobacterium B. subtilis strain KSAR2 may suppress the growth of R. solani and serve as an effective biocontrol agent for safeguarding plants from fungal infections. Additionally, diisooctyl phthalate could act as a natural, bioactive fungicide by inhibiting chitin synthase in *R. solani* and possibly

other fungi. This may ultimately result in significant progress in creating new and effective antifungal agents. Nonetheless, additional isolation and purification of these compounds, along with more practical studies for open-field applications, are necessary.

Conflicts of Interest: The author declares no conflicts of interest.

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