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Research Article

Geology**ASSESSMENT OF RAW MATERIALS FOR CEMENT INDUSTRY IN EL SUKHUNA-ZAAFRANA AREA, EASTERN DESERT, EGYPT**

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KEY WORDS

Cement Industry; Raw Materials; Red Kaolinitic clay; Gebel El Ramliya Limestone; Mineral Resources, Qiseib Formation, Abu Darag Kaolinitic Clay, Zaafrana Area, Eastern Desert, Egypt

ABSTRACT

In the last few years, cement industry becomes a major player in materials recycling, although it is environmental polluted industry and a great consumer of limestone and clay. Cement manufacturing depends greatly upon chemical and physical characteristics of raw materials, its quality and quantity, simplicity of its extraction and its costs. Ain El Sukhna area is surrounded by the Eocene limestone plateaus from the eastern and western sides. The plateaus are unconformity over the Permo-Triassic of Qiseib Formation that are made mainly of clays. The present study is aiming to preliminary assess the suitability and characterize raw materials (limestone and clay) in the western part of Zaafrana area for cement industry. Samples of limestone and clays were collected and analyzed for their main oxides. Geological and geochemical investigations indicated the suitability of the limestone of Sannor Formation and the Qiseib Formation kaolinitic clay for the cement industry in the area with considerable reserve.

Introduction

Portland cement is a hydraulic powder material, which reacts with water to produce strength-bearing lattices. The mixture of aggregates, cement and water is concrete. The strength and durability of concrete makes it one of the most useful materials developed by man. In simple terms, cement is a manmade mineral structure created at high temperatures, mainly comprising lime (CaO), Silica (SiO₂) and oxides of aluminum and iron (Al₂O₃ and Fe₂O₃). The cement making process can be divided; first 'clinker' is made at temperatures of 1400°C (Fig.1), then the clinker is milled with other minerals to produce the powder we know as cement (Kosmatka et al., 2002).

The raw materials used to produce clinker include some naturally occurring minerals and some materials available as waste streams from other industries. The most common combination of ingredients is limestone (for calcium) coupled with much smaller quantities of clay, shale and sand (as a source of silica, aluminum and iron). Other 'alternative' raw materials such as mill scale, fly ash and slag are brought in from other industries. Most cement plants are normally located on limestone deposits and shale or clay reserves to mine this locally. Additives are usually brought in, such as sands, iron ore and/or mill scales. Raw materials used in Portland cement manufacturing must contain appropriate proportion of calcium oxides, silica, alumina and iron oxides. According to Hewlett and Peter (1997) and Taylor (1997), the optimum chemical composition of clinker would have 67% CaO, 22% SiO₂, 5% Al₂O₃ and 3% Fe₂O₃. The relation between the four oxides has a direct impact on the clinker properties and its mineralogical compounds. Fortunately, different types of limestones and clays of different ages cover considerable parts of Egypt. In the North Eastern Portion of Egypt near the Red Sea coast, the carbonates usually form large extensive plateau that extending from Cairo to Red Sea coast.

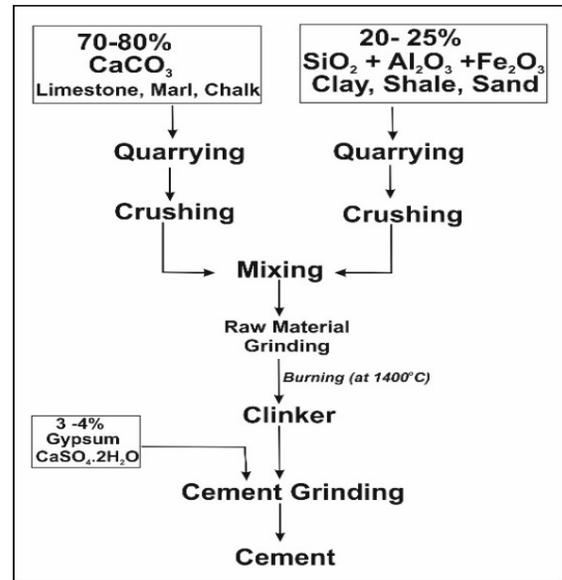


Fig.1: The life cycle of the cement production

Arabian Cement Company (ACC) plant is located at 29° 47' 43''N, 32° 08' 33''E, north of El Sukhna old road. This plant is located at the foot slope of Gebel El Ramliya limestone (Fig.2). At Gebel El Ramliya, Middle Eocene Mokattam group consists of Sannor (Observatory) Formation shallow marine dense medium bedded limestone with local chert and numulites.

In the area east of El Sukhna, large extension of limestone forming the cap rock of the plateau exists on both sides of the Cairo-Suez Road. The plateau extends from east of El Qattamiya to Red Sea. It extends to around 100 km east west and almost 200 km north east. The limestone plateau in the area is dissected by number of wadis. These wadis are; W. Gharaba, and W. Hagul. The existence of these wadis makes it possible for limestone quarrying and sampling due to the height of plateau and its steep slope (Fig.2).

Path from Reliance Heavy Industries (RHI) Clay Quarry (Abu Darag Clay Quarry) to Arabian Cement Company Plant is approximately 80 km. The Red Kaolinitic Clay deposits are of economic significance in the Abu Darag area. These kaolinitic clay deposits are a part of the

Paleozoic-Mesozoic sedimentary succession to the north-eastern part of Wadi Araba, along the southern slopes of the Northern Galala Plateau. The kaolinitic claystones of the Permo-Triassic (Qiseib Formation) are mostly sandy and alternate with sandstones in a cyclic pattern (Fig.

3). Clay in the area is overlain by an average of 3.5 m of sand and gravels (wadi deposits) and outcrops in few localities in the area before it covered by the reclaimed lands.



Fig. 2: ETM+ image of the study area showing the location of the studied sections: 1 Gebel Abu Treifiya, 2 Gebel Akheider, 3 Gebel El Ramliya, 4 Gebel El Qattamiya, 5 Gebel Umm Reheiat, 6 Wadi Gharaba, 7 Gebel Abu Shama, and 8 Bir Gindaly.

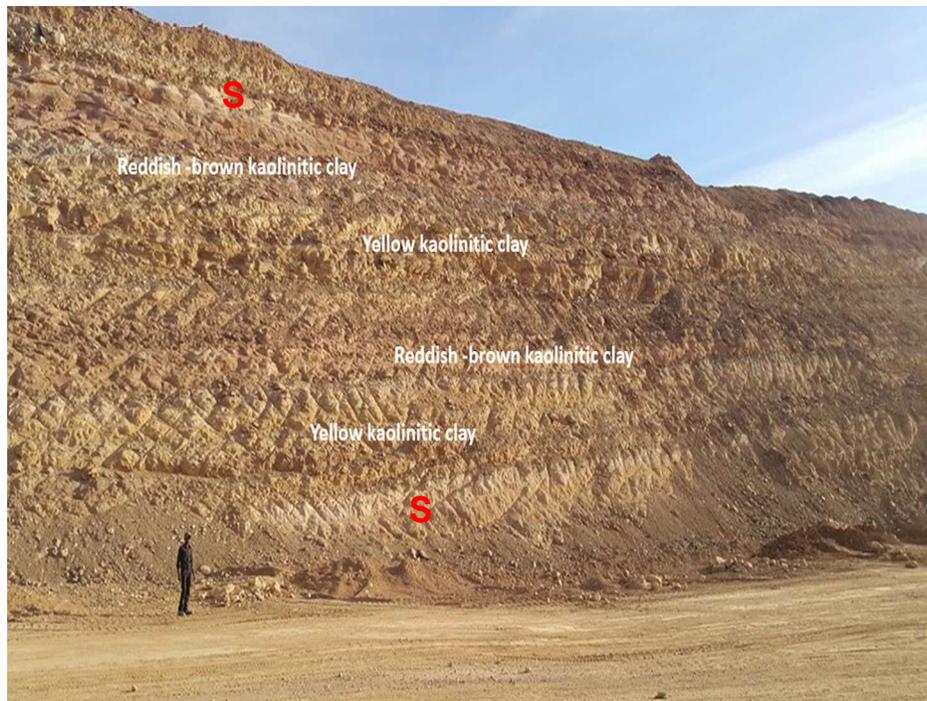


Fig.3: Intercalation flat- bedding variegated kaolinitic clay and sandstone in section II, Abu Darag Quarry, Red Sea area. **Notice:** upward gradation from thick to thin sandstone beds (S) that occur between the kaolinitic clay; Lower horizon: Yellowish–white kaolinitic clay; and Upper horizon: Reddish–Brown kaolinitic clay.

In June 2013 RHI (Reliance Heavy Industries) has signed a contract with the Egyptian Military – Al-Kheidma Al-Watania Sector. The contract is giving RHI the authority to extract the kaolinitic clays from an area of 1700m x 1000m with minimum rate of supply of 50,000 m³ per month.

The clay deposit lies at the western coast of the Gulf of Suez. The clay occurrence is an almost flat lying sequence of clay, sandy clay, and sandstone that occupies the low lands south of Gabal El Galala El Baharia (Fig. 4).

The main valleys and tributaries that traversed the area had masked appreciable parts of the clay exposures.

In 2005 core drilling was performed for this area in 29 core holes (See figure 5) from the top surface layer to varying depth depending on the continuity of the clay succession in the subsurface. The sequence penetrated is an alternation of red kaolinitic clay beds and sandy clay beds with frequent

sandstone intercalations (Fig. 3). The red kaolinitic clay is massive, indurated, dense, and light to greyish- brown in colour. The clay of this quality is semi-consolidated, kaolinitic, detrital, fine grained and dry. The sand is poor to well sorted, ranging in size from very fine to coarse sand, and even fine pebble size. The colors of the clay and sand varieties vary depending on the iron oxide contents

These natural resources are not utilized for any forms of industries and therefore the aim of this research is to highlight the chemical composition of the existing resources (limestone and clay) and to investigate their suitability for cement industries and make it available for investors to develop the area for job creation and make use of such raw materials. Also the study concerns with the assessment of the raw materials reserve to guarantee the sustainability for manufacturing. Small-scale private quarries of clay and sand are exploited in a very small-scale for building stones.

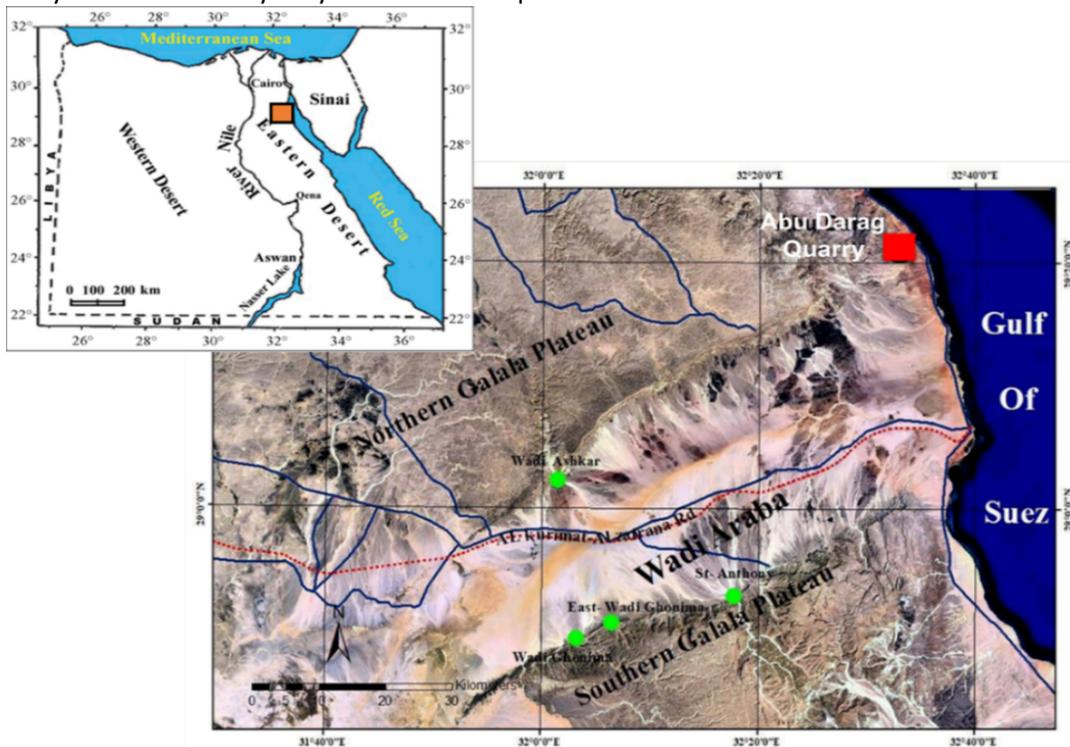


Fig.4: ETM+ image of the study area showing the location of the Wadi Araba area. The rectangular indicated to study Abu El Darag Clay Quarry area.

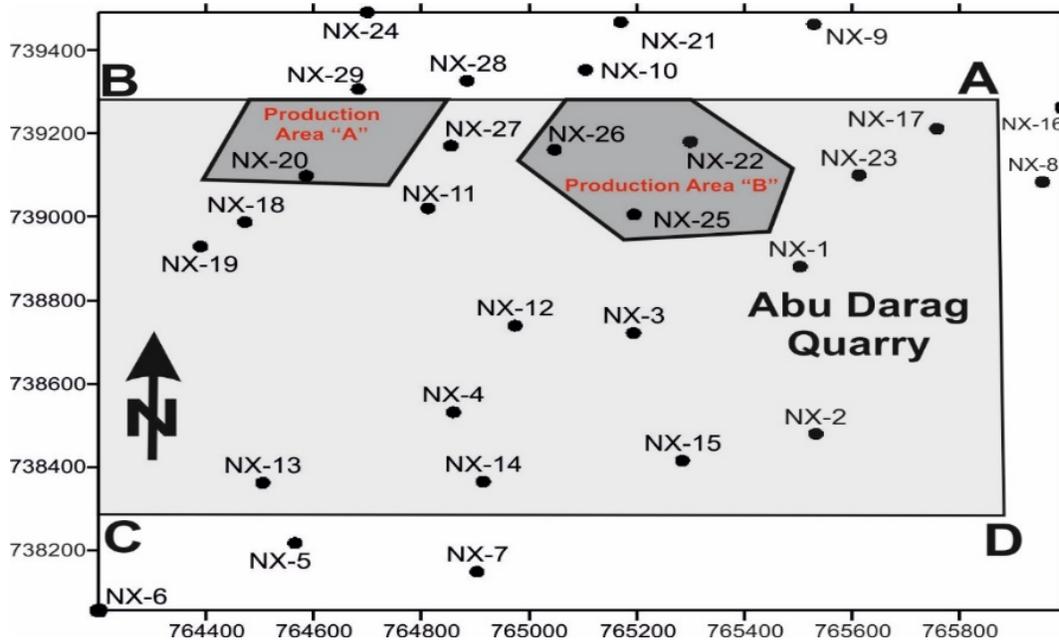


Fig.5: Areal distribution and location of 29 wells drilled in 2005 at Abu Darag area, Red Sea illustrations the Outlines boundaries of Abu Darag Quarry (ABCD); as well as production Areas "A" and "B".

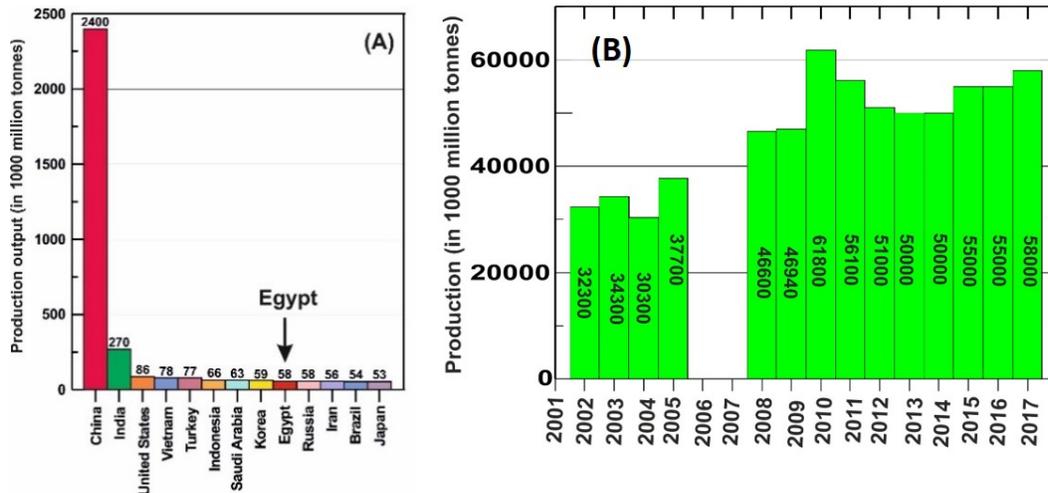


Fig.6: (A) Top Cement Producers in 2017; and (B) Egyptian cement production between 2001 and 2017 (in 1000 million tonnes).

International Cement Industry

About 4,100,000 Mt (million metric tons) of cement were producing in 2017 by 13 countries (Fig.6A). Asian countries were the highest of global producer cement with 75% of the global productions by China; India; Vietnam; Korea; Indonesia; Turkey; Saudi Arabia; and Iran

(Fig.6A). China is the 1st largest cement producer accounted for 2,400,000 Mt then India in the 2nd position (270,000 Mt). United States is the 3rd largest producers (86300 million tons).

Egypt's cement production is estimated to be 1.4 % (58,000 Mt) of the total world production 4,100,000 Mt (year 2017). On the other hand, Egypt is the largest cement producer

in Africa and occupied 9th position (58,000 Mt, Fig.6B) as well as consumer in the world led by the enormous growth in the infrastructure and construction sector for the last two decades.

Cement Manufacturing in Egypt

The cement industry in Egypt started in the twentieth century. In 1927, the Tourah Portland Cement Company (TPCC) was established, followed by the Helwan Cement Company (HCC) south of Cairo in 1929. Later in 1948, Alexandria Cement Company (APCC) was established followed by National Cement Company (NCC) in 1956. In the 70s, the production capacities of the 4 cement companies reached around 4 Mt (million metric tonnes) (ECS, 2009). The construction boom witnessed in the late 70s and 80s created high demand for cement that was met through imports because of the limited local production capacities, despite the opening of 3 new cement companies, Suez, Assuit and Amiryah, which started production throughout that period. In the mid-80s, Egypt became one of the largest cements importing countries in the world. During the 90s, 6 new cement companies were established to cope with the increasing construction activity and the resulting increasing cement demand, especially with the appearance of new sub-urban cities such as, Al-Sherouq, Al-Obour, 6th of October, 10th of Ramadan and Al-Sadat. Robust growth in the 1990s. Cement imports highest in 1999.

However, Egypt's cement net importing position prevailed during the 90s and the early years of the 21st century. Consequently, cement producers in Egypt increased their production capacities and enhanced their production lines to meet the surging local cement demand. In 2002, Egypt turned out to be a net exporter of cement and later in 2004 Egypt stopped importing cement and became one of the largest cement exporting countries in the region. Economic growth phase was slow during 2001/2002. Cement consumption declined by 4% during the year with 10.8 Mt of oversupply in 2003. Major national projects in Toshka, Suez Gulf and Sinai led to price war between

producers trying to gain or defend existing market share. New entrants attracted to industry.

Since the establishment of the Tourah Portland Cement Company (TPCC) in 1927 the Egyptian cement industry (ECI) has witnessed a steady growth. Also, a future look at the developments shows that huge expansions are to be expected. By establishing new or expanding existing production lines or by establishing new cement companies during the next three years Egypt aims to cope with the increasing consumption of the local market. It is estimated that around 16 Mt of clinker will be added, pushing the design capacity to 54.8 Mt of clinker. As for cement, it is expected that about 18 Mt will be added to the design capacity, which will increase from 46.6 Mt in 2008 to around 61.8 Mt in 2010. In 2017, the country's cement industry produced around 58 Mt of this building material. Furthermore, Egypt became one of the main exporting countries, with exports reaching their peak in 2017 at about 58 Mt of cement (Fig.6A). This statistic represents the volume of cement that was produced in Egypt from 2008 through 2017 (Fig.6B). Currently, the installed capacity of cement industry in Egypt reached approximately 62 Mt of cement per year, primarily from dry kilns, with only a small amount from seven wet kilns in two companies (Table 1; Fig.7). The satellite image in Fig.5 shows the locations of the 16 cement factories in Egypt.

Table 1 lists the production of each of the Egyptian cement-producing companies; all data refer to the year 2017, and in tandem with these data, the whole dataset has been updated with the latest figures by visiting a few Egyptian plants and doing an on-site survey due to the difficulty of acquiring statistics from a few plants because of breaching confidentiality in data acquisition.

Table 1: Cement company's clinker production in Egypt.

Group Name	Company Name	Facility Name	City	Klin Line*	Clinker Production (Million tonne)
Italcementi Group {Suez Cement Group of Companies (SCGC)}	Helwan Cement Company (HCC)	Helwan	Helwan	Dry Kiln 1 (plant 2), Dry Kiln 2 (plant 2), wet Kiln 2 (plant 1), wet Kiln 3 (plant 1), wet Kiln 5 (plant 1), wet Kiln 6 (plant 1), wet Kiln 1 (plant 3), wet Kiln 2 (plant 3)	4,009,340
	Tourah Portland Cement Company (TPCC)	Tourah	Tourah	Kiln 5, Kiln 7, Kiln 8, Kiln 9	2,474,412
	Suez Cement Company (QCP)	Qattamya Plant	Qattamya	Kiln 1	9,845,810
	Suez Cement Company (SCC)	Suez Plant	Suez	Kiln 1, Kiln 2	2,100,710
	Suez Cement Company	El Minya Plant SCC	El Minya	Kiln 1	287,666
	National Cement Company (NCC)	Al-Tebbin	South Helwan	Kiln 1 (wet), Kiln 2 (wet), Kiln 3, Kiln 4	3,031,951
	El Sewedy Cement Company (ESCC)	Ataqah-Suez	Ataqah	Kiln 1	
Lafarge Holcim Ltd	Lafarge Cement Egypt, S.A.E. (LCC)	Suez	Suez	Kiln 1, Kiln 2, Kiln 3, Kiln 4, Kiln 5	8,295,478
Cementos La Union, S.A.	Arabian Cement Company (ACC)	Ain El Sukhuna	Suez	Kiln 1	2,030,428
Cemex	Cemex Egypt (Assiut Cement)	Assiut	Assiut	Kiln 1, Kiln 2, Kiln 3	4,706,112
	Wadi El Nile Cement Company (WNCC)	Wadi El Nile Cement Company	Wadi El Ebyari/ Ashyab		
	El Naha Cement	Qena	Qena		
	Medcom Cement	Aswan Cement Plant	Aswan		

	Royal El Minya Cement	El Minya	Samalut		
ASEC Cement	Misr Cement Qena Co.	ASEC Minya	Samalut		
		Qena	Nasr City	Production line	1,859.73
Helwan Cement Company	Helwan – El Menia White Cement Company	El Minya	El Minya		
Building Materials Industries Company (BMIC)	South Valley Cement S.A.E. (SVCC)	Beni Suef	Beni Suef	Kiln 1	1,573,844
		BMI Assisut Plant	Assiut		
Egyptian Armed Forces	El Arish Cement Company	El Arish	Beni Suef	Cement plant with six production lines in the Egyptian city of Beni Suef	
			El Arish	Kiln 1, Kiln 2	
Titan Group	Titan Cement Egypt	Beni Suef	Beni Suef	Kiln 1	1,573,844
	Alexandria Portland Cement Company (APCC)	El Mex	Alexandria	Kiln 1	1,500,005
Cimentos de Portugal, SGPS, S.A. (Cimpor)	Amreyah Cement Company	Amreyah	Burg al Arab	Kiln 1, Kiln 2	1,900,483
Aalborg Portland A/S	Sinai White Portland Cement Co.	Sinai White	El Arish		
Vicat Group	Sinai Cement Company	Sinai	El Arish	Kiln 1	

* In terms of the kiln process type, primarily from dry kilns, with only a small amount from seven wet kilns in two companies (National cement and Helwan (Italcementi group).



Fig.7: Distribution of cement factories in Egypt are highlighted with red indicators

Raw Materials Sampling and Reserve Estimation

According to the Egyptian Geological Survey reports, the thickness of limestone in the eastern side of Gebel Ataqa reaches 200 m or more. This thickness extends laterally for a long distance. This limestone becomes slightly dolomitic at the lower part and chalky at upper part. Arabian Cement Company (ACC) plant is located at $29^{\circ} 47' 43''$ N, $32^{\circ} 08' 33''$ E, north of El Sukhna old road. This plant is located at the foot slope of Gebel El Ramliya limestone. At Gebel El Ramliya, Middle Eocene Mokattam group consisting of Sannor (Observatory) Formation shallow marine dense medium bedded limestone with local chert and numulites. Field observation and measurements indicate that the average thickness of the limestone can be quarried ranges between 45 m and 70 m (Sannor Formation). About 200 Million

tonne of proven limestones can be quarrying for 50 years

Clays crop-out at several localities, mainly reddish-brown, grey to yellowish-grey, brown colour in parts and sandy in others. The thickness of these clays varies from place to another. In some places its thickness ranges from 5 m to 23 m of clay and silty clay with overburden reach to 3 m (EGSMA, 1988). Samples were collected from the study area for chemical analysis. Fifty-five samples of limestone in five areas in the southeastern part of the province were collected in the form of nine trenches. Twenty-two clay samples representing the Abu Darag areas were collected from quarries in the Abu Darag area (Fig.4). These samples were analyzed chemically by XRF and the physical properties were

assessed. Table 2 shows the preferred range of some parameters in the raw materials for cement industries as given by Taylor (1997). Table 3 shows the chemical analysis of the 55 limestone samples collected from the Gabel El Ramlaya area, while the results of the chemical analysis of the 22 clay samples collected from Abu Darag area is given in Table 4. On the other hand, Table 5 illustrates the average chemical analysis of these analyzed samples of limestone and clay.

Quality of the Raw Materials

Portland cement is made by powder heating of batches mixture of limestone and clay and/or other materials up to a temperature of $\sim 1450^{\circ}\text{C}$. Partial fusion occurs and nodules or balls of clinker are produced. The clinker is ground mixture with a 5% of raw gypsum to control rate of thin the produced cement. The cement clinker typically has a composition in the region of 67% CaO, 22% SiO₂, 5% Al₂O₃, 3% Fe₂O₃ as well as 3% of other components, and normally contains four major phases; alite, belite, Ca-aluminates and ferrite phases (Kosmatka et al., 2002). Other phases, such as

alkali sulphates and calcium oxide are normally present in minor amounts (Taylor, 1964).

Cement experts are care for some factors that determine quality of the raw materials. The most important factors are the Lime Saturation Factor (LSF), the Silicate Ratio (SM) and the Aluminum Ratio (AM) (Cement Sustainability Production Report, 2011).

The LSF is used for proportioning the raw mix. It is the ratio of CaO to the other three main oxides and is calculated as: $LSF = \text{CaO} / (2.8\text{SiO}_2 + 1.18\text{Al}_2\text{O}_3 + 0.65\text{Fe}_2\text{O}_3)$ and varies between 0.88 and 0.95 for Portland cement. Typical LSF values in modern clinkers are 0.92-0.98. Values above 1.0 indicate that free lime is likely to be present in the clinker. This is because, in principle, at LSF=1.0 all the free lime should have combined with belite (2CaO.SiO₂) to form alite (3CaO.SiO₂). If the LSF is higher than 1.0, the surplus free lime has nothing with which to combine and will remain as free lime. In practice, the mixing of raw materials is never perfect and there are always regions within the clinker where the LSF is locally a little above, or a little below, the target for the clinker as a whole. This means that there is almost always some residual free lime, even where the LSF is considerably below 1.0. It also means that to convert virtually all the belite to alite, an LSF slightly above 1.0 is needed.

Table 2: Preferred range of some parameters in raw material (after Taylor, 1997).

Item	Range (%)	Preferred	Effects
K ₂ O, Na ₂ O	0 – 1	0.2 – 0.3	Improve burnability at lower temperature and deteriorates at higher temperature, creates operational problems
SO ₃ ⁻²	0 – 4	0.5 – 2	When S > (Na, K) acts as modifier of alkali recycle by forming less volatile (Na, K) SO ₄ compounds. 2.5–4% causes sulphate expansion
Cl ⁻¹	0–0.6	0.0–0.015	Increases ring formation, chloride in concrete promotes corrosion of steel
MgO	0 – 5	0 – 2	Reduce viscosity of liquid phase, quicker formation of C ₃ S then decomposition into C ₂ S
LSF	88–95		
SM	>3		
AM	>2	1.6–1.8	

Table 3: Chemical analysis of limestone from Sannor Formation at Gebel El Ramliya

Time	SmpGrp	SiO ₂	Al ₂ O ₃	Fe ₂ O ₃	MgO	CaO	Na ₂ O	K ₂ O	SO ₃	Cl	LOI	CaCO ₃	LSF	SM	AM
5/1/2018 0:00	Crush_Lst	0.38	0.12	0.05	0.86	53.74	0.09	0.01	0.30	0.03	43.18	95.13	4341	2.24	2.40
5/1/2018 18:51	LSLOADING	0.70	0.28	0.22	0.49	51.95	0.08	0.02	0.24	0.04	41.36	91.96	2135	1.40	1.27
5/1/2018 18:51	LSLOADING	0.15	0.09	0.04	0.59	54.00	0.09	0.01	0.06	0.05	43.09	95.59	9779	1.15	2.25
5/2/2018 0:00	Crush_Lst	0.16	0.09	0.04	0.92	54.44	0.07	0.01	0.09	0.02	43.79	96.37	9383	1.23	2.25
5/3/2018 12:00	Crush_Lst	0.53	0.31	0.15	0.37	52.91	0.06	0.01	0.05	0.02	41.98	93.66	2717	1.15	2.07
5/3/2018 13:30	LSLOADING	0.53	0.31	0.15	0.36	52.92	0.06	0.01	0.05	0.02	41.99	93.68	2718	1.15	2.07
5/4/2018 0:00	Crush_Lst	2.11	1.05	0.59	0.70	47.48	0.15	0.04	0.07	0.07	38.08	84.05	631	1.29	1.78
5/5/2018 12:00	Crush_Lst	0.62	0.17	0.07	1.04	52.17	0.17	0.05	0.10	0.09	42.12	92.35	2632	2.58	2.43
5/6/2018 0:00	Crush_Lst	0.35	0.19	0.52	0.54	54.35	0.08	0.02	0.13	0.04	43.29	96.21	3524	0.49	0.37
5/6/2018 12:24	LSLOADING	0.07	0.06	0.03	0.20	56.00	0.05	0.01	0.07	0.02	44.24	99.13	19560	0.78	2.00
5/6/2018 12:24	LSLOADING	0.09	0.06	0.03	2.69	51.41	0.11	0.01	0.06	0.06	43.34	91.00	15019	1.00	2.00
5/7/2018 0:00	Crush_Lst	0.05	0.06	0.06	7.67	43.04	0.24	0.01	0.12	0.09	42.17	76.19	17230	0.42	1.00
5/7/2018 12:00	Crush_Lst	0.21	0.10	0.05	0.21	55.45	0.06	0.02	0.11	0.02	43.81	98.16	7508	1.40	2.00
5/7/2018 12:29	LSLOADING	0.20	0.10	0.05	0.20	55.47	0.06	0.02	0.11	0.02	43.81	98.19	7807	1.33	2.00
5/7/2018 12:29	LSLOADING	0.06	0.06	0.02	0.62	55.38	0.06	0.01	0.09	0.01	44.21	98.03	21994	0.75	3.00
5/7/2018 12:29	LSLOADING	0.27	0.09	0.06	0.46	54.70	0.07	0.02	0.11	0.04	43.49	96.83	6070	1.80	1.50
5/7/2018 12:29	LSLOADING	0.06	0.05	0.08	8.18	43.07	0.20	0.01	0.10	0.08	42.76	76.24	15437	0.46	0.63
5/8/2018 0:00	Crush_Lst	1.03	0.47	0.28	0.13	52.36	0.07	0.02	0.10	0.02	41.30	92.69	1446	1.37	1.68
5/8/2018 14:15	LSLOADING	0.11	0.06	0.05	1.25	53.30	0.08	0.01	0.06	0.04	43.26	94.35	12959	1.00	1.20
5/8/2018 14:16	LSLOADING	0.50	0.15	0.11	2.87	49.00	0.15	0.02	0.08	0.10	41.63	86.74	2972	1.92	1.36
5/10/2018 0:00	Crush_Lst	0.53	0.21	0.07	1.41	52.40	0.14	0.02	0.05	0.12	42.73	92.76	2948	1.89	3.00
5/11/2018 12:00	Crush_Lst	1.44	0.37	0.24	0.16	50.80	0.15	0.08	0.14	0.03	40.10	89.92	1098	2.36	1.54
5/12/2018 17:41	LSLOADING	0.54	0.29	0.18	0.36	53.07	0.05	0.02	0.06	0.02	42.10	93.94	2692	1.15	1.61
5/12/2018 17:41	LSLOADING	0.10	0.09	0.05	0.80	54.13	0.05	0.01	0.08	0.01	43.42	95.82	12928	0.71	1.80
5/13/2018 0:00	Crush_Lst	0.08	0.07	0.03	1.27	53.93	0.11	0.01	0.05	0.08	43.75	95.46	16538	0.80	2.33
5/14/2018 13:24	LSLOADING	0.39	0.14	0.08	0.15	54.51	0.08	0.02	0.08	0.02	43.00	96.49	4164	1.77	1.75
5/14/2018 13:24	LSLOADING	0.22	0.11	0.05	0.40	54.26	0.07	0.02	0.08	0.02	43.09	96.05	6972	1.38	2.20
5/14/2018 13:24	LSLOADING	0.10	0.07	0.03	0.13	55.25	0.05	0.01	0.06	0.01	43.57	97.80	14460	1.00	2.33

Table 3 (Cont.)

Time	SmpGrp	SiO ₂	Al ₂ O ₃	Fe ₂ O ₃	MgO	CaO	Na ₂ O	K ₂ O	SO ₃	Cl	LOI	CaCO ₃	LSF	SM	AM
5/16/2018 12:00	Crush_Lst	0.18	0.09	0.04	1.68	52.26	0.05	0.01	0.17	0.01	42.91	92.51	8214	1.38	2.25
5/16/2018 14:33	LSLOADING	0.17	0.08	0.04	1.68	52.26	0.05	0.01	0.17	0.01	42.91	92.51	8763	1.42	2.00
5/17/2018 0:00	Crush_Lst	0.19	0.10	0.05	1.69	52.79	0.10	0.01	0.15	0.06	43.34	93.45	7735	1.27	2.00
5/17/2018 13:23	LSLOADING	0.52	0.29	0.19	0.15	53.69	0.03	0.02	0.03	0.01	42.37	95.04	2794	1.08	1.53
5/17/2018 13:24	LSLOADING	0.10	0.09	0.03	0.08	55.56	0.03	0.01	0.03	0.00	43.76	98.35	13695	0.83	3.00
5/17/2018 13:24	LSLOADING	0.40	0.12	0.09	0.12	54.21	0.05	0.02	0.15	0.03	42.74	95.96	4107	1.90	1.33
5/18/2018 0:00	Crush_Lst	0.36	0.13	0.06	2.76	49.81	0.14	0.02	0.48	0.08	42.16	88.17	4149	1.89	2.17
5/18/2018 12:00	Crush_Lst	0.30	0.12	0.06	0.69	53.00	0.04	0.02	0.14	0.01	42.39	93.82	5193	1.67	2.00
5/19/2018 0:00	Crush_Lst	0.13	0.08	0.03	0.35	55.51	0.06	0.01	0.10	0.02	44.01	98.26	11615	1.18	2.67
5/19/2018 12:00	Crush_Lst	0.10	0.07	0.05	1.47	53.76	0.05	0.01	0.19	0.02	43.83	95.16	13607	0.83	1.40
5/20/2018 12:00	Crush_Lst	0.42	0.14	0.10	0.15	54.41	0.06	0.02	0.36	0.03	42.94	96.31	3869	1.75	1.40
5/20/2018 14:37	LSLOADING	0.36	0.17	0.09	0.42	54.67	0.05	0.02	0.08	0.01	43.43	96.77	4315	1.38	1.89
5/20/2018 14:37	LSLOADING	0.42	0.14	0.10	0.15	54.38	0.07	0.02	0.35	0.03	42.91	96.26	3867	1.75	1.40
5/22/2018 14:31	LSLOADING	0.23	0.11	0.09	0.94	54.13	0.05	0.01	0.12	0.02	43.57	95.82	6504	1.15	1.22
5/22/2018 14:31	LSLOADING	0.20	0.09	0.09	1.04	53.88	0.05	0.01	0.07	0.03	43.48	95.38	7435	1.11	1.00
5/22/2018 14:31	LSLOADING	0.20	0.09	0.08	0.37	55.25	0.05	0.01	0.09	0.02	43.83	97.80	7693	1.18	1.13
5/23/2018 0:00	Crush_Lst	0.20	0.07	0.06	0.46	55.13	0.04	0.01	0.12	0.02	43.83	97.59	8088	1.54	1.17
5/24/2018 12:00	Crush_Lst	0.27	0.10	0.06	0.67	54.34	0.07	0.02	0.10	0.03	43.44	96.19	5952	1.69	1.67
5/25/2018 0:00	Crush_Lst	0.22	0.09	0.05	0.95	54.14	0.08	0.01	0.04	0.06	43.58	95.84	7174	1.57	1.80
5/26/2018 0:00	Crush_Lst	0.85	0.30	0.22	1.20	50.65	0.09	0.03	0.53	0.05	41.11	89.66	1761	1.63	1.36
5/27/2018 0:00	Crush_Lst	1.32	0.47	0.28	0.64	50.81	0.08	0.03	0.07	0.01	40.65	89.94	1146	1.76	1.68
5/27/2018 12:00	Crush_Lst	0.31	0.10	0.08	0.96	52.95	0.07	0.02	0.13	0.02	42.66	93.73	5101	1.72	1.25
5/27/2018 12:37	LSLOADING	0.09	0.06	0.04	1.01	53.54	0.05	0.01	0.13	0.01	43.18	94.77	15350	0.90	1.50
5/27/2018 12:37	LSLOADING	0.30	0.10	0.08	0.96	52.94	0.07	0.02	0.13	0.02	42.65	93.71	5242	1.67	1.25
5/29/2018 15:42	LSLOADING	0.31	0.13	0.10	1.26	52.52	0.09	0.03	0.05	0.02	42.65	92.97	4834	1.35	1.30
5/29/2018 15:42	LSLOADING	1.58	0.47	0.18	1.19	48.46	0.23	0.11	0.10	0.03	39.38	85.78	951	2.43	2.61
5/30/2018 0:00	Crush_Lst	0.22	0.09	0.06	0.67	54.21	0.06	0.02	0.24	0.03	43.34	95.96	7122	1.47	1.50

$$\text{Lime Saturation Factor (LSF)} = \text{CaO} / (2.8\text{SiO}_2 + 1.18\text{Al}_2\text{O}_3 + 0.65\text{Fe}_2\text{O}_3)$$

$$\text{Silicate Ratio (SM)} = \text{SiO}_2 / (\text{Al}_2\text{O}_3 + \text{Fe}_2\text{O}_3).$$

$$\text{Aluminum Ratio (AM)} = (\text{Al}_2\text{O}_3 / (\text{Fe}_2\text{O}_3))$$

Table 4: Chemical analysis of kaolinitic clay from Abu Darag area.

Date	SiO ₂	Al ₂ O ₃	Fe ₂ O ₃	MgO	CaO	Na ₂ O	K ₂ O	SO ₃	Cl	LOI	LSF	SM	AM
Tuesday, May 1, 2018	62.36	14.14	5.02	0.27	0.32	0.19	0.60	1.70	0.24	14.69	0.16	3.26	2.82
Wednesday, May 2, 2018	60.01	17.78	7.50	0.35	0.31	0.17	0.67	0.09	0.19	12.44	0.16	2.37	2.37
Thursday, May 3, 2018	59.34	17.65	7.73	0.23	0.31	0.16	0.60	0.19	0.18	13.12	0.16	2.34	2.28
Friday, May 4, 2018	60.07	16.50	7.83	0.31	0.31	0.13	0.51	0.19	0.17	13.49	0.16	2.47	2.11
Saturday, May 5, 2018	61.02	16.61	6.47	0.23	0.31	0.15	0.54	0.21	0.18	13.76	0.16	2.64	2.57
Sunday, May 6, 2018	62.85	15.59	6.77	0.20	0.31	0.23	0.74	0.13	0.10	12.62	0.15	2.81	2.30
Monday, May 7, 2018	62.24	15.78	7.22	0.25	0.30	0.13	0.64	0.04	0.14	12.60	0.15	2.71	2.19
Tuesday, May 8, 2018	64.34	13.53	7.45	0.17	0.31	0.11	0.58	0.11	0.07	12.88	0.15	3.07	1.82
Wednesday, May 9, 2018	64.20	14.23	6.67	0.32	0.31	0.15	0.68	0.06	0.06	12.87	0.15	3.07	2.13
Thursday, May 10, 2018	63.27	15.45	7.23	0.16	0.30	0.20	0.60	0.04	0.06	12.26	0.15	2.79	2.14
Friday, May 11, 2018	64.27	14.00	6.50	0.19	0.30	0.16	0.76	0.48	0.17	12.75	0.15	3.14	2.15
Saturday, May 12, 2018	61.54	14.92	7.67	0.18	0.31	0.12	0.62	0.07	0.14	13.95	0.16	2.72	1.94
Sunday, May 13, 2018	60.49	15.45	6.28	0.26	0.31	0.14	0.59	0.31	0.14	15.54	0.16	2.78	2.46
Monday, May 14, 2018	64.29	12.84	5.93	0.28	0.31	0.11	0.46	0.13	0.12	15.05	0.16	3.43	2.17
Tuesday, May 15, 2018	60.03	16.96	8.58	0.27	0.31	0.18	0.50	0.05	0.10	12.51	0.16	2.35	1.98
Wednesday, May 16, 2018	61.03	17.60	8.09	0.22	0.30	0.17	0.66	0.02	0.11	11.30	0.15	2.37	2.17
Friday, May 18, 2018	60.30	18.34	6.80	0.21	0.30	0.12	0.44	0.04	0.09	12.92	0.15	2.40	2.70
Saturday, May 19, 2018	63.78	14.74	5.73	0.26	0.31	0.13	0.60	0.43	0.10	13.43	0.15	3.12	2.57
Sunday, May 20, 2018	62.60	17.35	5.97	0.23	0.31	0.18	0.61	0.12	0.09	12.05	0.15	2.68	2.90
Monday, May 21, 2018	61.53	19.89	6.99	0.28	0.31	0.10	0.63	0.21	0.10	9.42	0.15	2.29	2.84
Tuesday, May 22, 2018	63.16	19.75	6.47	0.22	0.31	0.22	0.56	0.12	0.09	8.64	0.15	2.41	3.05
Sunday, May 27, 2018	61.53	19.89	6.99	0.28	0.31	0.10	0.63	0.21	0.10	9.42	0.15	2.29	2.84
Monday, May 28, 2018	63.16	19.75	6.47	0.22	0.31	0.22	0.56	0.12	0.09	8.64	0.15	2.41	3.05
Tuesday, May 29, 2018	57.14	22.97	9.25	0.22	0.30	0.12	0.60	0.02	0.05	8.85	0.16	1.77	2.48
Wednesday, May 30, 2018	57.07	22.91	7.93	0.23	0.31	0.10	0.57	0.02	0.06	10.31	0.16	1.85	2.89
Thursday, May 31, 2018	59.90	21.40	8.25	0.22	0.30	0.15	0.58	0.05	0.08	8.59	0.15	2.02	2.59

$$\text{Lime Saturation Factor (LSF)} = \text{CaO} / (2.8\text{SiO}_2 + 1.18\text{Al}_2\text{O}_3 + 0.65\text{Fe}_2\text{O}_3)$$

$$\text{Silicate Ratio (SM)} = \text{SiO}_2 / (\text{Al}_2\text{O}_3 + \text{Fe}_2\text{O}_3).$$

$$\text{Aluminum Ratio (AM)} = \text{Al}_2\text{O}_3 / (\text{Fe}_2\text{O}_3)$$

The silica ratio (also known as the Silica Modulus) is defined as: $SM = SiO_2 / (Al_2O_3 + Fe_2O_3)$. A high silica ratio means that more calcium silicates are present in the clinker and less celite ($3CaO \cdot Al_2O_3$) and ferrite ($3CaO \cdot Al_2O_3 \cdot Fe_2O_3$). SM exits ≤ 3 for most cement products and is typically between 2.0 and 2.8.

The alumina ratio (also known as the Alumina Modulus) is defined as: $AM = (Al_2O_3 / (Fe_2O_3))$. This determines the potential relative proportions of celite ($3CaO \cdot Al_2O_3$) and ferrite ($3CaO \cdot Al_2O_3 \cdot Fe_2O_3$) phase in the clinker. It is ≤ 2 and the preferable range should be between 1.6 and 1.8 because lower values leads to lower clinkering temperature. Moreover, an increase in clinker AM means there will be proportionally

more celite ($3CaO \cdot Al_2O_3$) and less ferrite in the clinker.

The other controlling factors are the level of the MgO, SO₃ and Cl. If higher levels of SO₃ and Cl exit in the raw materials, troubles in the kiln during burning are expected. The MgO content should not exceed 5% in raw materials as it forms small crystals of periclase which are slowly hydrated in the cement or concrete expand and cause cracking. Clays usually contain limited amounts of Na₂O, K₂O, SO₃ and Cl <0.3, 2% and 0.015% respectively, it would be suitable for manufacturing cement, while silica must not exceed three times the percentages of Al₂O₃ + Fe₂O₃ as previously mentioned by Youssef (1994).

Table 5: The average chemical analysis of limestone and clay samples in the studied area.

	Limestone (Gebel El Ramliya)				Kaolinitic Clay (Abu Darag)			
	Min	Max	Mean (n=55)	Std	Min	Max	Mean (n=26)	Std
SiO ₂	0.05	2.11	0.39	0.41	57.07	64.34	61.60	2.04
Al ₂ O ₃	0.05	1.05	0.17	0.16	12.84	22.97	17.15	2.80
Fe ₂ O ₃	0.02	0.59	0.11	0.11	5.02	9.25	7.07	0.96
MgO	0.08	8.18	1.07	1.49	0.16	0.35	0.24	0.05
CaO	43.04	56.00	52.92	2.66	0.30	0.32	0.31	0.00
Na ₂ O	0.03	0.24	0.08	0.05	0.10	0.23	0.15	0.04
K ₂ O	0.01	0.11	0.02	0.02	0.44	0.76	0.60	0.07
SO ₃	0.03	0.53	0.13	0.10	0.02	1.70	0.20	0.33
Cl	0.00	0.12	0.03	0.03	0.05	0.24	0.12	0.05
LOI	38.08	44.24	42.76	1.20	8.59	15.54	12.08	2.06
CaCO ₃	76.19	99.13	93.68	4.70				
LSF	631	21994	7344	5226	0.15	0.16	0.16	---
SM	0.42	2.58	1.37	0.48	1.77	3.43	2.60	0.42
AM	0.37	3.00	1.77	0.57	1.82	3.05	2.44	0.36

$$\text{Lime Saturation Factor (LSF)} = CaO / (2.8SiO_2 + 1.18Al_2O_3 + 0.65Fe_2O_3)$$

$$\text{Silicate Ratio (SM)} = SiO_2 / (Al_2O_3 + Fe_2O_3).$$

$$\text{Aluminum Ratio (AM)} = (Al_2O_3 / (Fe_2O_3))$$

Quantitative Chemical Analysis of the Limestone and Clay in the Area

Quantitative chemical analysis of the limestone samples

The main component of limestone is the calcium carbonates (CaCO_3), which on heating dissociates into CaO and CO_2 . According to the data of Table 5 the average range of silica falls between 0.05% and 2.11%; these contents are relatively high in some outcrops. The average ranges of alumina, iron oxide, lime and magnesia contents fall between 0.05–1.05%, 0.02–0.59%, 43.04–56.00%, and 0.08–8.18%, respectively. The high level of magnesia oxide (MgO) is related to the dolomitic bands in the lower part of the Mokattam group which consisting of observatory formation shallow marine dense medium bedded limestone with local chert and numulites. Obviously, the MgO content in some samples exceeds 5% in raw materials as it forms small crystals of periclase which are slowly hydrated in the cement or concrete expand and cause cracking. The high MgO content also led to reduce viscosity of liquid phase, quicker formation of Tricalcium Silicate (Alite phase: C_3S) then decomposition into Belite (C_2S). The average ranges of alkalis, chloride and sulphate contents are ranging between 0.04 and 0.34%, 0.03%, and 0.13%; respectively. The average range of lime saturation factor (LSF) is between 631% and 21994%. While silica modulus (SM) is ranging between 0.42% and 2.58% and Alumina modulus (AM) between 0.37% and 3.00%. The low silicate is mainly owing to the low contents of silica and low contents of alumina and iron oxides. The average total carbonate is between 76.19% and 99.13%, while the non-carbonate constituents are around 1–15%. The loss on ignition falls between 38.08% and 44.24%. Accordingly, based on the above-mentioned factors and their ranges, the limestone in Gabel El Ramliya area are suitable for the cement industry due to the presence of low sulphate and chloride.

Quantitative chemical analysis of the Abu Darag kaolinitic clay samples

According to the data shown in Tables 4 and 5, the average range for contents of silica falls at (57.07 and 64.34%), alumina (12.84 and 22.97%), iron oxides (5.02 and 9.25%), lime (~0.30%), magnesia (0.16 and 0.35%), alkalis (0.56 and 0.97%), chlorides (0.05 and 0.24%) and sulphates (0.02 and 1.70%). The calculated average range of lime saturation factor (LSF) is between 0.15% and 0.16%, these values are very low due to increasing silica, alumina and iron oxides contents. The average range of Silica Modulus (SM) exists between 1.77% and 3.43% and Alumina Modulus (AM) ranges between 1.82% and 3.05% and the loss on ignition falls between 8.59% and 15.54%. For clay, the proportion of silica to iron oxide and alumina oxide ranges around 1.77 to 343 (average 2.60; within the recommended range), which are adequate for cement industry.

The chemistry of the Raw Mix in the Study Area

Cements are generally combinations of limestone, shells or chalk, and shale, clay, sand or iron ore, usually mined from a quarry close to the plant. The basic chemical components of Portland Cement (PC) are calcium (Ca), silicon (Si), aluminum (Al) and iron (Fe). The percent of limestone and clay forming raw material mix varies from plant to another according to their material available, plant specifications or KPI's. By the way, cement raw materials percent are trials that meet some standard. It is usual that the limestone component is around 75% of the mix and the clay forms ~25%. Assuming that, at a round of previous rates of limestone and the clay are to be used for raw materials, it is supposed to be the limits of the important and influential elements, depending on the average chemical analysis of clay and limestone on the area as follows: the average lime content should be range between 38% and 42%, while magnesia content falls between 0.68% and 3.55%. The MgO will cause some operational problems if higher MgO content of limestone is used as

corrective materials. The average alkalis ($K_2O + Na_2O$) content ranges between 0.42% and 1.13%. The average chloride content lies between 0.15% and 0.71%. The average sulphate content ranges between 0.12% and 0.52%. The average lime saturation factor ranges between 81% and 110%. The average silica modulus is ranging between 1.78% and 2.84%; low silica modulus can be increased using corrective sands. The average alumina modulus falls between 1.22% and 1.94%. Homogeneity of limestone is preferred to overcome the higher content of silica and magnesia.

Conclusion

The study area is located in the northern portion of the Eastern Desert of Egypt in the El Sukhna-Zaafrana old road on the Red Sea.

The foot slope of Gebel El Ramliya limestone. At Gebel El Ramliya, Middle Eocene Mokattam group consisting of Sannor (Observatory) Formation shallow marine dense medium bedded limestone with local chert and numulites. The average thickness of the limestone can be quarried ranging between 45 m and 70 m (Sannor Formation). In the area east of El Sukhna, large extension of limestone forming the cap rock of the plateau exists on both sides of the Cairo-Suez Road. The plateau extends from east of El Qattamiya to Red Sea, to around 100 km east west and almost 200 km north east. The limestone plateau in the area is dissected by number of wadis. The limestone plateau is made of massive carbonates with high content of chert bands. These wadis are W. Gharaba, and W. Hagul. The existence of these wadis makes it possible for limestone quarrying and sampling due to the height of plateau and its steep slope.

The limestone plateaus unconformably lies over the Permo-Triassic of Qiseib Formation that are made mainly of clays and kaolinitic clays. The sequence penetrated is an alternation of kaolinitic clay beds and sandy clay beds with frequent sandstone intercalations. The kaolinitic clay is massive, indurated, dense, and light to greyish-brown in colour. The clay of this quality is semi-consolidated, kaolinitic, detrital, fine grained and dry. The sand is poor to well sorted, ranging in size from very fine to coarse sand, and even fine pebble size. The colors of the clay and sand varieties vary depending on the iron oxide contents.

The chemical analyses of 55 samples of limestone and 22 samples of clays that is made to evaluate their suitability for cement industries indicate that:

- Abu Darag red kaolinitic clay (RKC) in the area is suitable for cement industry. Where about 41 Million tonne of proven Abu Darag red kaolinitic clay (RKC) can be quarried for 30 years
- Field observation and measurements indicate that the average thickness of the limestone can be quarried ranging between 45 m and 70 m (Sannor Formation). About 200 Million tonne of proven limestones can be quarried for 50 years.
- The limestone of Sannor (Observatory) Formation is not suitable for cement industries with regard to the high content of chert bands and silica concretions.
- The limestone of Drunka Formation especially the upper most parts (50–70 m) satisfies all the requirements for the manufacturing of the Portland cement clinker.

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تقييم المواد الخام لتصنيع الأسمنت بمنطقة السخنة – الزعفران، الصحراء الشرقية – مصر

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الطين هو المادة الخام التي لديها العديد من الاستخدامات عبر الصناعات المختلفة. استخدام الرئيسي من الطين في منطقة أبو الدرج – قصب هو عمل الاسمنت. وتهدف الدراسة إلى تحديد خصائص رواسب الطين الأحمر بمنطقة أبو الدرج اعتماداً على الخصائص المعدنية والجيوكيميائية والفيزيائية. وتظهر مكاشف رواسب الطين الأحمر من البرمو- ترياسي (تكوين قصب)، في منطقة أبو الدرج- قصب حيث تمتد موازية لخليج السويس على مساحة حوالي 50 كم² مع حوالي 54 مليون طن احتياطي. ويتكون الطين الأحمر أساساً من الكاولين و الكوارتز وكميات ثانوية من السميكتيت (S)، الأليت (I) وطبقات مختلطة من I/S و الهيماتيت. ويتميز الطين الأحمر باحتوائه على نسبة قليلة لكل من الكبريتات (SO₃ = 0.35%) و الأكاسيد القلوية (Na₂O = 1.34% و K₂O = 1.06%). ويتميز Clinkers بقيم قابلة للاسترداد منخفضة لكل من SO₃ (0.13%), Na₂O (0.42%) و K₂O(0.38%) وكذلك يتميز بانخفاض قليلاً في سليكات ثلاثي الكالسيوم (طور alite) و ألومنيات ثلاثي الكالسيوم (طور aluminat) ولكن بزيادة في بيتا سليكات ثنائي الكالسيوم (طور belite) و ألومنيات حديد رباعي الكالسيوم (طور ferrite). وبالمقارنة مع الأسمنت البورتلاندي العادي، فمعاجين الإسمنت من الطين الأحمر التي قيد دراسته تتميز بما يلي: (1) وقت طويل من التميؤ والإعداد، (2) طاقة التميؤ الحرارية المتولده منخفضة، (3) قوة عالية تتطور بعد 28 يوماً، (4) عالية المقاومة لتأثيرات المواد الكيميائية. وفي ضوء النتائج التي تم التوصل إليها فإنه يستنتج أن الطين الأحمر من تكوين قصب يمكن استخدامه لإنتاج أنواع خاصة من الأسمنت، مثل الاسمنت المقاوم للمواد الكيميائية والأسمنت المنخفض الحرارة.