PHYTOCHEMICAL STUDY OF EUPHORBIA HETEROPHYLLA L. CULTIVATED IN EGYPT

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ABSTRACT

A phytochemical study of Euphorbia heterophylla L. cultivated in Egypt is presented. β -amyrin, β -sitosterol taraxasterol acetate, β -sitosterol glucoside and hydrocarbon with ketonic group were isolated from pet. ether extract of the herb. While lupeol acetate, taraxasterol, β -sitosterol and β -sitosterol glucoside were isolated from ether extract of the root. In addition quercetin, 3-methyl quercetin, kaempferol-3-0-arabinoside, kaempferol-3-0-glucoside and kaempferol-7-0-glucoside were isolated from the ethyl acetate extract of the herb. The identification of these compounds was based on physical, chemical and spectral analysis.

INTRODUCTION

Euphorbia heterophylla L. is an annual herb helonging to family Euphorbiaceae. It was used as an antidote for the irritation produced by other species of Euphorbia. The flower and leaf gave positive antibiotic tests against T.B. 1. The aqueous extract of the leaves was investigated for the purgative effect in animals possibly due to the increase in intestinal motility 2. The plants of the genus Euphorbia have been reported to contain terpenoids (di- and tri-), alcohol, sterols, hydrocarbons and flavonoids.

On the other hand several other substances, viz, alkaloids, coumarins, tannins and acids were reported 3-7.

Current literature on *E. heterophylla* L. revealed the presence of triterpenes euphyl acetate and mortenone as well as 10,10-dimethylhexacosane-7-one 8. The present work is directed for studying the lipids, terpenes and flavonoids of this plant.

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EXPERIMENTAL

The plant material consisting of the aerial parts and roots collected during the flowering stage from the Experimental Station of Medicinal Plants, Faculty of Pharmacy, Assiut University. The plant was identified through "Garden Plants of the World in Colours", and kindly confirmed by Prof. Dr. N.E. El-Keltawi, Professor of Horticulture, Faculty of Agriculture Assiut University.

Melting points were determined using a koffler hot stage microscope. ¹H-NMR spectra were recorded in CDCl₃. CDCl₃+pyridine-d₅ and DMSO-d₆ at 400 MHz from Bruker WH 400. Mass spectra were measured using MS-50, Kratos, A.E.I. 70 ev. Unicam infra-red spectrophotometer SP-1025 for recording infrared spectra and Unicam SP-1750 recording Ultra-violet spectrophotometer for UV measurements.

Extraction:

a-The powdered herb of *E. heterophylla* L. (3 Kg) was extracted several times with 70% ethanol by percolation till exhaustion. The dried alcoholic extract (150 g) was mixed with 600 ml warm distilled water, and fractionated into pet. ether (Fr. I) and ethyl acetate (Fr. II). The pet. ether fraction was evaporated, the residue (70 g) was extracted with methanol (500 ml) (Fr. Ia) and then evaporated to yield a brownish residue (20 g) (Fr. Ib).

b-The powdered roots (700 g) was extracted with 70% ethanol as mentioned before. The dried alcoholic extract (25 g) was mixed with 200 ml warm dis-

tilled water and partitioned between ether (Fr. III) and n-butanol.

Investigation of the Pet. Ether Fraction of the Powdered herb:

A few spots of (Fr. Ia) was chromatographed on silica gel-coated plates using benzene-ethyl acetate (9:1) system I and chloroform-methanol (9:1) system II. The chromatograms were sprayed with 50% H₂SO₄, followed by heating at 110°C for few minutes. Eight spots were obtained having R_f values 0.96, 0.85, 0.74, 0.55, 0.38, by system I and 0.99, 0.98, 0.97, 0.87, 0.80, 0.40, 0.36 and 0.24 by system II.

Column Chromatography of the Pet. Ether Fraction of the Powdered herb:

The methanol-soluble part (20 g) of (Fr. Ib) was transferred to a silica gel column (E-Merck, 500 g, 1 m x 4.5 cm, i.d.) and gradiently eluted with hexane and ethyl acetate. Fractions, 500 ml each were collected and monitored by silica gel G. coated plates using system I and II. Identical fractions were pooled together. Five compounds labelled E₁, E₂, E₃, E₄, and E₅ were obtained.

Compound E₁:

Compound E₁ (20 mg) occurred as white waxy substance, ethyl acetate, m.p. 74-76°C, R_f 0.97 & 0.99 by systems I & II respectively. It did not respond to Salkowski's and Liebermann-Burchard's tests. IR spectrum (KBr) showed the following bands 2950, 1710, 1450 and 1380 cm⁻¹. ¹H-NMR (CDCl₃) showed signal at 50.88 (m, for terminal CH₃, groups) 1.27, 1.59, 1.62 and

2.41 (m. for CH₂ protons). MS showed M⁺ at m/z 408, diagnostic peaks at 390, 362 and other peaks spaced 14 mass units (corresponding to a difference of CH₂).

From the aforementioned chemical, physical and spectral data of compound E₁ it can be concluded that it is a hydrocarbon containing a ketonic group. It is most probably 10, 10-dimethyl hexacosane-2-one, previously isolated from the plant under investigation 8.9.

Compound E2:

Compound E₂ (1 g) is colourless hexagonal plates, (ethyl-acetate) m.p. 248-50°C. It responded to Salkow-ski's and Liebermann-Burchard's tests. It had R_f values 6.74 and 6.97 by systems I & II respectively. IR (KBr \checkmark) showed the following bands 2966. 1725, 1460, 1375, 1250 and 1030 cm⁻¹. ¹H-NMR (CDCl₃) showed six singlets at 5 6.73, 0.844, 0.849, 0.86, 0.90 and 0.93 ppm (6, s. 3H each, for 6 Me): 0.94 (3H, d, J=6.1 Hz, Me-29); 2.04 (3H, s, OAc); 1.01-2.3 (m, for methylene protons): 4.5 (1H, dd, J=9.9, 6.7 Hz, H-3) 5.2 (2H, t like, J=3.3 Hz, for methylene protons at C-39).

It was identified as taraxasterol acetate from the previous mentioned physicochemical and spectral properties ¹⁰ as well as direct comparison (mmp & Co-chromatography) with an authentic sample.

Compound E3:

Compound Eg (80 mg) is colourless needles. m.p. 198-200°C (methanol). It responded to Salkowski's and Liebermann-Burchard's tests. It had Rf values 0.55 and 0.87 by systems I & II respectively. It was identified as β -amyrin by IR, co-chromatography and mmp with and authentic sample.

Compound E4:

Compound E4 (800 mg) is colourless needles, m.p. $135-37^{\circ}$ C (methanol). It reponded to Salkowski's and Liebermann-Burchard's tests. It had Rf values 0.38, 0.80 by systems I & II respectively. It was identified as β -sitosterol by IR, co-chromatography and mmp with a reference material.

Compound E5:

Compound Es (50 mg) is white amorphous powder, (MeOII) m.p. 275-8°C. It responded to Salkowski's and Liebermann-Burchard's tests. It had Rf value 0.24 by system II. IR spectrum of the compound showed bands at 3400. 2940, 1450, 1370, 1610, 1070, and 1020 cm⁻¹. ^{1}H -NMF spectrum (400 MHz, CDCl3+pyridine-d5) showed the following signals at & 0.67 ppm (S, Me-18) & 0.92 ppm (S, Me-13) & 0.92 ppm (d, J=6.1 Hz, Me-21) & 0.82 ppm (d. J=6.9 Hz. Me-26) & 0.84 ppm (d. J=6.9 Hz. Me-27) & 0.84 ppm (t, J=7.3 Hz, Me-29) & 3.55 ppm (m, H-3) & 5.3 ppm (t, J=5.01 Hz, H-6) & 1-2.5 ppm (m, CH₂ and CH protons) & 4.58 ppm (d, J=7.7 Hz, H-1') & 3.77 ppm (t, J=8.8 Hz. H-2',3',4') & 3.65 ppm (m, H-5') & 4.07 ppm (dd, J=11.5, 3.5 Hz, H-6'-a) & 3.94 ppm (dd, J=11.4, 5.4 Hz, H-6'b). Ms showed a molecular ion peak M⁺ at m/z 414 and other diagnostic peaks at 399(M⁺-Ne), 396(M^+-H_2O) 100%, 382(M^+-Me-H_2O), 294, 275(M-R), 255, 241, 213, 171, 161, 145, 133, 121, 107, 95, 81, 73, 69, 67, 57.

From the aforementioned physicochemical and spectral studies the compound E5 was identified as β -sitosterol glucoside according to the published data for β -sitosterol 11,12. This was confirmed by completed acid

hydrolysis of the compound and the chromatographic study of both the aglycone and the sugar.

Investigation of the Ethyl Acetate Fraction (Fr. II) of the Powdered herb:

TLC investigation of (Fr. II) was performed using the following systems:

-For silica gel G plates:

System III: ethyl acetate-formic acid-water (10:2:3).

System IV: chloroform-MeOH (8:2).

-For cellulose plates:

System V: chloroform-acetic acid-water (50:45:5).

The chromatograms showed 7 flavonoidal compounds that were revealed by UV, ammonia vapour and 1% AlCl₃ spraying reagent.

Column Chromatography:

The ethyl acetate fraction (Fr. II) (15 g) was chromatographed on silica gel column (E-Merck, 300 g. 120X2.5 cm) and eluted with chloroform then chloroformmethanol gradient. Fractions 250 ml each collected and monitored by TLC and PC and pooled according to similar Rf. Six groups were obtained and five flavonoids labelled F_1 , F_2 , F_3 , F_4 , and F_5 were isolated.

Compound F₁:

It was obtained as yellowish powder (MeOH) (50 mg), m.p. $315-17^{\circ}$ C. It had R_f values 0.98, 0.63 by systems III & IV successively. From the UV spectropho-

tometric data with shift reagents Table 1, Co-chromatography and mmp with a reference sample it was concluded that compound F1 is quercetin.

Compound F₂:

Compound F₂ (80 mg) was obtained as yellow needles, m.p. 258-70°C (methanol). It had R_f values 0.98, 0.50 by systems III & IV. The UV spectrophotometric data were recorded in Table 1. 1 H-NMR (400 MHz, DNSOd6) showed the following signals at δ 3.78 ppm (3H, S, 0CH3 at C₃) & δ .19 ppm (1H, d, J=1.9 Hz, H-6) & δ .41 ppm (1H, d, J=1.7 Hz, H-8) & δ .90 ppm (1H, d, J=3.5, H-5°) & 7.45 ppm (1H, dd, J=2.1 and 8.5 Hz, H-6°) 7.54 ppm (1H, d, J=2.1 Hz, H-2°).

Compound F_2 was identified as 3-methyl quercetin.

Compound F3:

It is yellowish powder, (39 mg), m.p. 275°C (decomp.) (methanol). It had Rf values 0.90, 0.87 by system III and V. The UV spectrophotometric data Table 1 proved that this substance is a flavonol glycoside. Partial and complete acid hydrolysis indicated that it is a monoside giving kaempferol aglycone (AF3) that its UV spectrophotometric data were recorded in Table 1. The sugar was identified as arabinose. From the aforementioned data compound F3 was found to be kaempferol-3-0-arabinoside.

Compound F₄:

It was obtained as yellowish powder (30 mg) and decomposed at 265°C (methanol). It had Rf values 0.85, 0.81 by systems II, V. The UV spectrophotometric data Table 1 proved that this compound is a flavonoi glycoside.

Partial and complete acid hydrolysis indicated that it is a monoside giving kaempferol aglycone on complete acid hydrolysis.

The sugar moiety was identified as glucose. From the above mentioned data compound F_4 was identified as kaempferol-7-V-glucoside.

Compound F5:

Compound F_5 was obtained as yellowish powder (60 mg) m.p. 242-44°C (methanoi). It had R_f values 0.81. 0.75 by systems III, V. The UV spectrophotometric data were recorded in Table 1. It shows that this substance is flavonol glycoside.

Partial and complete acid hydrolysis of this substance indicated that it is a monoside giving kaempferol aglycone on complete acid hydrolysis.

The sugar moiety was identified as glucose. From aforementioned data compound F5 was identified as kaempferol-3-0-glucoside.

Investigation of Ether Fraction (Fr. III) of Powdered Roots:

TLC on silica gel G of (Fr. III) using benzeneethyi acetate (9.5:0.5) system VI and system I showed 8 spots that were visualized by spraying with 50% $\rm H_2SO_4$ followed by heating at 110°C. The spots had $\rm R_f$ values 0.98, 0.96, 0.95, 0.91, 0.88, 0.85, 0.80 and 0.24 by system II & 0.97, 0.95, 0.78, 0.60, 0.67, 0.44 and 0.31 by system VI.

Column Chromatography:

Fr. III (7 g) was chromatographed on silica gel (E.Merck, 280 g, 1 m X 2.5 cm) using hexane and hexaneethyl acetate gradient. Fractions 500 ml each were collected and monitored by TLC using systems II and VI. They were pooled according to similar $R_{\rm f}$. Six groups were obtained and four compounds labelled $R_{\rm l}$, $R_{\rm l}$, $R_{\rm l}$, $R_{\rm l}$, and $R_{\rm l}$ were isolated.

Compound R₁:

Colourless needles (600) m.p. 211-13'C (methanol). It responded to Salkowski's and Liebermann-Burchard's tests. It had R_f values 0.96 and 0.99 by systems II & VI respectively. IR spectrum showed bands at 2930, 1730, 1640, 1460, 1380, 1250 and 1130 cm⁻¹. ¹H-NMR spectrum (CDCl₃) showed six singlets at δ 0.73, 0.85, 0.36, 0.93, 0.94 and 1.01 ppm (each for Me group) & 1.63 ppm (3H, s, C=C-CH₃), 2.04 ppm (3H, s, O-C-CH₃) & 4.46 ppm (2H, m, CH₂) & 5.2 ppm (2H, broad singlet, H-3). MS showed a molecular ion peak M⁺ at m/z 468, peaks at 453(M⁺-Me), 408(M⁺-HOAc), 393(M⁺-HOAc-Me) and other peaks at 257, 249, 231, 218, 204(100%), 189, 177, 161, 147, 135, 121, 109, 95, 84, 81, 69, 66, 43, 28. This pattern of fragmentation is characteristic for lupane series ¹³.

The aforementioned physicochemical and spectral data of this compound superimpose those reported for lupeol acetate 14 .

Compound R2:

Colourless needles (800 mg), m.p. 223-25°C (methanol). It responded to Salkowski's and Lieber-mann-Burchard's tests. It had Rf values 0.85 and 0.44

by systems II & VI respectively. IR spectrum showed hands at 3420. 2950, 1610, 1470, 1390, 1190. 1040 and 980 cm $^{-1}$. 1 H-NMR (400 MHz, CDCl₃) showed six singlets at δ 0.79, 0.37, 0.88, 0.95, 0.99 and 1.04 ppm (each for Me group) & 1.05 ppm (3H, d, Me at C₂₉) & 3.19 ppm (1H, dd, J=10.5, 6.5 Hz, H-3) & 4.60 ppm (2H, dd, methylene protons at C-30). MS showed a molecular ion peak M⁺ at m/z 426 and other diagnostic peaks at 411 (M⁺-Me), 408 (H⁺-H₂O), 393 (M⁺-H₂O-Me), 357, 344, 315. 272, 257, 229, 218, 307, 189 (100%), 175, 161, 147, 135, 121 and 109.

The above mentioned data superimpose those reported for taraxasterol 16 .

Acetylation of compound R_2 gave compound E_2 that was isolated from the pet. ether fraction of the powdered herb.

Compounds R₃ and R₄ were proved to be β -sitosterol and β -sitosterol glucoside following the same procedure mentioned under E₄ & E₅.

RESULTS AND DISCUSSION

From the aerial parts of Euphorbia heterophylla L. cultivated in Egypt, β -amyrin, β -sitosterol, β -sitosterol glucoside taraxasterol acetate and 10,10-dimethyl hexacosane-2-one were isolated from pet. ether extract. In addition quercetin, 3-methyl quercetin, kaempferol-3-0-arabinoside, kaempferol-3-0-glucoside and kaempferol-7-0-glucoside were isolated from ethyl acetate extract of the herb, while lupeol acetate, taraxasterol, β -sitosterol and β -sitosterol glucoside were isolated from the ether extract of the root.

The identity of the isolated compounds was confirmed through determining their physical and chemical characters, as well as their chromatographic and spectral analysis.

Referring to the literature euphyl acetate, 10,10-dimethyl hexacosane-7-one and mortenone were previously isolated from *E. heterophylla* L. while the other compounds are reported here for the first time.

Table 1: UV-spectra of isolated flavonoids F 1 F 2 F 3 F 4 F 5 and AF 3.

Compound	Band	MeOH	Į	nax		and			Max				B0 ₃
					₩		AlCl ₃ /HCl		NaOAc		NaOAc/Hg		
		254 370											·
F2	II I				276 434			+20 +48		+18 +2 0		+18	
F3		264 350		+24 +64		+4 + 4 8		+4 +48		+4 +10		-	
F ₄	II I	266 346	282 398	+16 +52	27 0 356	+4+10	27 0 356	+4+10	265 346	-	266 346	-	
F5		264 350			270 398			+6 +48		+4 +8		-	
·AF3		268 366						+2 +58		+6 +8			

Phytochemical Study of Euphorbia heterophylla L. Cultivated in Egypt

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دراسة الكيمياء العقاقيرية لنبات اليوقوروبيا هيتيروقيللا ل. المنزرع فى ممصر نصر نصر احمد العمرى – مقبول احمد مقبول – محمد عبد المطلب عبد الحافظ سلوى قاروق قرج

قسم العقاقير - كالية الصيدلة - جامعة استيوط

من خلاصة البترول الایثیری لمسحوق اعشاب النبات تم فصل والتعرف علی خمسة مرکبات هی ۱۰و۰۱-دای میثیل هکساکوزان-۲-اون ، خلات التراکساستیرول ، بیتا سیتوستیرول وبیتا سیتوستیرول جلوکوزید.

ومن خلاصة خلات الایشیل لمسحوق الاعشاب تم ایضا فصل والتعرف عملی کویرستین ۳-مثیل کویرستین ، کامبیوفرول-۳-i-ارابینوزید ، کامبیوفرول-۳-i-جلوکوزید.

كذلك تم فصل والتعرف على خلات الليبيول ، تراكساستيرول ، بيتا سيتوسيترول ، بيتا سيتوسيترول جلوكوزيد من خلاصة الاثير لجذور النبات.

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