BIOACTIVE PRINCIPLES OF THE FLOWERS OF PANCRATIUM MARITIMUM

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تم في هذا البحث تجزئة وتنقية جزء خلات الايثيل من الخلاصة الكحولية للأزهار الطازجة لنبات البانكريشيم مارتيمم المنزرع في مصر وقد أسفرت هذه الدراسة عن فصل قلوانى البانكراتستاتين بالاضافة الى - كافويل كوينك ميثيل استر وقد تم تعيين التركيب الكيميائى لهذه المركبات باستخدام الرنين النووى المغناطيسي البروتوني والكربوني أحادي وثنائي الاحداثيات وكذلك استخدام مطياف الكتلة عالى التحلل وقد أعطى كلا المركبين فعالية جيدة ضد الخلايا السرطانية المسماه هيلا وفعالية متوسطة كمضاد للدرن ضد ميكوبكتريم تيوبركلوزس وتم لأول مرة فصل قلوانى البانكراتستاتين من البانكريشيم ماريتمم أما المركب - كافويل كوينك ميثيل استر فهذه أول مرة يفصل فيها من الفصيله النرجسية

Bioactive-guided fractionation and purification of the ethyl acetate fraction of an ethanolic extract of fresh flowers of Pancratium maritimum (Amaryllidaceae) cultivated in Egypt yielded an alkaloid, pancratistatin (1), together with 3caffeoylquinic acid methyl ester (2). Their structures were determined on the basis of extensive 1D (¹H and ¹³C) and 2D (COSY, HMQC, HMBC, and NOESY) NMR studies and high-resolution mass spectral measurements. Both compounds showed potent cytotoxic activity against HeLa cells and moderate anti-tuberculosis activity against Mycobacterium tuberculosis H37Rv. This is the first report of pancratistatin in Pancratium maritimum and the first report of 3-caffeoylquinic acid methyl ester in the family Amaryllidaceae.

INTRODUCTION

In continuation of our on going investigation and search for drug leads from the Egyptian Amaryllidaceae plants,¹⁻⁶ we have investigated the fresh flowers of *Pancratium maritimum* and found that the ethyl acetate fraction of an ethanolic extract of the fresh flowers possesses cytotoxic activity against HeLa cells. Bioactive-guided fractionation and purification of the active fractions furnished two potent cytotoxic compounds namely pancratistatin (1) and 3-caffeoylquinic acid methyl ester (2) (Figure 1).

The structural determination of the isolated compounds was established using different spectroscopic techniques including UV, IR, HRFABMS, and 1D (¹H and ¹³C) and

2D (¹H-¹H COSY, HMQC, HMBC, and NOESY) NMR studies.

Pancratistatin is reported for the first time from the Egyptian *Pancratium maritimum* and this is the first report of 3-caffeoylquinic acid methyl ester in the family Amaryllidaceae. In addition, the tuberculostatic activity of both compounds is reported here for the first time.

Previous studies on different parts of the plant in our laboratory resulted in the isolation of alkaloids, chromones, chalcones, acetophenones, flavans, and flavonoids.²⁻⁶

The present work deals with the isolation, structural mapping and the biological (cytotoxic and anti-tuberculosis) evaluation of compounds **1** and **2**.

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EXPERIMENTAL

General experimental procedures

Mps were uncorrected. Optical rotations were measured on JASCO DIP-1000 digital polarimeter. UV spectra were recorded on a Hitachi 300 spectrometer. ¹H and ¹³C NMR spectra were recorded on a JEOL α -500 spectrometer at 500 MHz for ¹H and 125 MHz for ¹³C, respectively. NMR chemical shifts were referenced to TMS as internal standard. Positive FAB mass spectral data were obtained with a JEOL JMS-700T mass spectrometer using 3-nitrobenzylalcohol (3-NBA) as matrix. HPLC was performed on preparative (ARII Cosmosil, 250 x 20 mm) and semipreparative (ARII Cosmosil, 250 x 10 mm) C18 columns (Waters) with a UV detector at 220 nm and flow rate of 5.5 or 2.0 mL/min. Precoated silica gel 60 F₂₅₄ plates (E. Merck) were used for TLC.

Plant material

The fresh flowers of *Pancratium maritimum* were collected in May 2000 from the cultivated plants at the campus of Suez Canal University. The plant material was authenticated by Prof. Dr. A. Fayed, Professor of plant taxonomy at Assiut University. A voucher specimen was deposited in herbarium of the Department of Pharmacognosy, at Faculty of Pharmacy, Suez Canal University under the registration No. PM1.

Extraction and isolation

The fresh flowers (6.85 kg) were crushed into small pieces and macerated in ethanol 70 % (3 x 10 L) at room temperature. The combined extracts were evaporated under reduced pressure. The brown viscous residue was dissolved in 500 mL water and was successively extracted with n-hexane (3 x 250 mL) (5.32 g), CH₂Cl₂ (3 x 250 mL) (2.45 g), ethyl acetate (3 x 250 mL) (3.55 g), and nbutanol (3 x 250 mL) (10.21 g). Preliminary cytotoxicity assay on the crude fractions showed that the ethyl acetate fraction possesses cytotoxicity against HeLa cells. The ethyl acetate residue (3.55 g) was subjected to flash chromatography on ODS column (40 x 3.0 cm) starting with 10% MeOH in H₂O through pure MeOH (each 500 mL). The fraction eluted with 20% MeOH showed moderate cytotoxicity $(IC_{50} = 1.8 \ \mu g/mL)$ against HeLa cells. The residue of this fraction (1.25 g) was flash chromatographed on ODS column (40 x 3.0 cm) starting with H₂O through pure MeOH (each 500 mL). The fractions eluted with 20% MeOH through 40% MeOH showed similar TLC patterns and were cytotoxic (IC₅₀ = 0.8µg/mL) to Hela cells. The residue of these fractions (520 mg) was flash-chromatographed on ODS column (30 x 2.0 cm) starting with 5% MeOH in H₂O through pure MeOH (each 300 mL). The fraction eluted with 15% MeOH was potent cytotoxic (IC₅₀ = $0.08 \ \mu g/mL$) to HeLa cells. This fraction (118 mg) was finally purified on a preparative C18 HPLC column (ARII Cosmosil, 250 x 20 mm, Waters) using 30% MeCN in H₂O to afford 2 (10.9 mg) and an impure 1 (48 mg). Compound 1 was further purified on a semi-preparative C18 HPLC column (ARII Cosmosil, 250 x 10 mm, Waters) using 25% MeOH in H₂O to afford 17.8 mg of

Compound (1): White solid. m.p 323-325°. $[\alpha]_D^{25} = +48.6^\circ$ (DMSO, c = 0.1). UV (MeOH) λ_{max} : 280, 237 nm. Positive HRFABMS: obsd m/z 326.0879 [M + H]⁺ (calcd for C₁₄H₁₆NO₈, 326.0876). IR(KBr) ν_{max} : 3419, 1672, 1629, 1468, 1358, 1083, 1047cm⁻¹. ¹H and ¹³C NMR data are listed in Table 1.

Compound (2): Yellowish oil. UV (MeOH) λ_{max} : 329, 298, 235, 219 nm. Positive HRFABMS: obsd m/z 369.1189 [M + H]⁺ (calcd for C₁₇H₂₁O₉, 369.1186). IR (KBr) ν_{max} : 3420, 1664, 1520, 1163, 1027, 625 cm⁻¹. ¹H and ¹³C NMR data are listed in Table 2.

RESULTS AND DISCUSSION

Compound 1

pure 1.

Compound 1 was purified as white solid. positive **HRFABMS** showed The pseudomolecular ion peak at m/z 326.0879, which is in consistence with the molecular formula of $C_{14}H_{16}NO_8$ [M + H]⁺. Its IR spectrum showed absorption bands at 3419 and 1672 cm⁻¹, which are characteristic for OH and amide moieties, respectively. The 500 MHz ¹H NMR spectrum $(DMSO-d_6)$ displayed resonances for 15 protons including six exchangeable singlets (Table 1). From the counter plot of the ¹H-¹H COSY and HMQC

No.	$\delta_{\rm C}$ (mult.)	$\delta_{\rm H}$ [mult., J(Hz)]	HMBC with H
1	70.20 (d)	3.95 (br.s.)	H-3, H-4a
<i>H</i> O-1		5.35 (br.s.)	
2	68.48 (d)	4.27 (br.s.)	H-10b
НО-2		4.82 (br.s.)	
3	73.29 (d)	3.83 (br.s.)	H-1
HO-3		5.06 (br.s.)*	
4	69.98 (d)	3.72 (br.s.)	H-2, N <i>H</i>
НО-4	—	5.06 (br.s.)*	
4a	50.51 (d)	3.71 (m)	
5	—	7.50 (s)	
6	169.48 (s)		NH, H-10
ба	107.48 (s)		NH, H-10, H-10b
7	145.35 (s)		NH, H-10
<i>H</i> O-7	—	13.04 (br.s.)	
8	131.69 (s)		H-10, OC <i>H</i> ₂ O
9	152.04 (s)		H-10, H-10b, OCH ₂ O
$\overline{OCH_2O}$	101.75 (t)	6.04 (s), 6.02 (s)	
10	97.67 (d)	6.47 (s)	H-10b
10a	135.66 (s)		H-4a, H-10b
10b	39.50 (d)	2.95 (d, 12.4)	H-1, H-4, H-10, NH

Table 1: ¹H and ¹³C chemical shift data of compound 1 (*DMSO-d*₆).

*) Overlapped signals.

Table 2: ¹H and ¹³C chemical shift data of compound **2** (CD_3OD).

No.	$\delta_{\rm C}$ (mult.)	$\delta_{\rm H}$ [mult., $J({\rm Hz})$]	HMBC with H	NOESY with H
1	75.31 (s)		H ₂ -2, H-3, H-5, H ₂ -6	
2	36.36 (t)	2.15 (dd, 14.3, 3.7)		
		2.05*		
3	72.60 (d)	5.30 (m)	H ₂ -2, H-5	H-5
4	73.83 (d)	3.63 (dd, 7.6, 3.0)	H ₂ -2, H-5, H ₂ -6	H-5
5	68.60 (d)	4.06 (ddd, 8.3, 8.3, 3.7)	H-3, H-4, H ₂ -6	H-3, H-4
6	40.80 (t)	2.03*		
		1.97 (dd, 13.4, 8.3)		
7	176.41 (s)		H ₂ -2, H ₂ -6, H ₃ -8	
8	52.85 (q)	3.67 (s)		
1'	127.91 (s)			
2'	115.09 (d)	6.99 (br.s.)		
3'	146.78 (s)		H-2', H-5', H-6'	
4'	149.45 (s)		H-2′, H-5′, H-6′	
5'	116.45 (d)	6.70 (d, 8.2)		
6'	122.90 (d)	6.89 (d, 8.2)		
7'	146.84 (d)	7.55 (d, 16.0)	H-8′	
8'	115.73 (d)	6.25 (d, 16.0)		
9'	168.91 (s)		H-7′, H-8′	

*) Signals are partially overlapped.



Fig. 1: Structures of the isolated compounds.

experiments, the existence of a single continuous spin-coupling system consisting of six resonating signals could be traced through the molecule of **1**. The coupling system starts from the resonating signal at δ 3.95 (H-1) with coupling to the signals at δ 2.95 (H-10b) and δ 4.27 (H-2), which further couples to H-3 at δ 3.83. Further coupling between H-3 and H-4 (δ 3.72), between H-4 and H-4a (δ 3.71), and finally between H-4a and H-10b was observed, completing the spin-coupling system. Further, COSY cross-peaks between H-1, H-2, H-3, H-4 and the exchangeable singlets at δ 5.34 (HO-1), 4.82 (HO-2), 5.06 (HO-3), and 5.06 (HO-4), respectively, were observed, confirming the tetrahydroxy substitution of the cyclic hexane ring of **1**.

In addition, the ¹H NMR spectrum showed two one-proton singlets at δ 6.04 and 6.02 for a methylenedioxy moiety and a singlet at δ 6.47 for H-10.

Furthermore, the exchangeable singlets at δ 13.04 and 7.50 correspond to the phenolic OH at C-7 and the NH functionalities, respectively.

The ¹³C NMR spectrum together with HMQC experiment revealed resonances for 14 carbons (Table 1) including six singlets, one triplet, and seven doublets. Unambiguous assignment of all protonated carbons was made

possible from HMQC experiment, while the unequivocal assignment as well as the connectivities of the three rings of **1** were confirmed by HMBC experiment (Table 1). For Example, cross-peaks of H-10/C-6, NH/C-6 (δ 169.48), as well as H-10/C-6a, H-10b/C-6a, and NH/C-6a (δ 107.48) confirmed the assignments of these quaternary carbons. Similarly, the assignment of C-7, C-8, C-9, and C-10a was established from HMBC correlations and are listed in Table 1.

The above-mentioned data are in good agreement with the data reported for pancratistatin isolated previously from the roots and bulbs of *Pancratium littorale*^{7,8} (later reassigned as *Hymenocallis littorale*).⁹

Compound 2

Compound 2 was purified as a yellowish oil. Its positive HRFABMS showed a pseudomolecular ion peak at m/z 369.1189 which is consistent to the molecular formula of $C_{17}H_{21}O_9$ [M + H]⁺. The IR spectrum showed absorption bands at 3420 and 1664 cm⁻¹ corresponding to OH and ester moieties, respectively. Its 500 MHz ¹H NMR (*CD*₃*OD*) spectrum revealed resonances for 15 protons (Table 2). Interpretation of the ¹H-¹H COSY and HMQC experiments showed the existence of three coupling systems. The first system includes the protons H-1/H₂-2/H-3/H-4/H-5/H₂-6. The second system (ABX system) is formed of the protons H-2', H-4', and H-6', while the trans-coupled protons H-7' and H-8' (J = 16.0Hz) constitutes the third coupling system (AB system) (Table 2).

Interpretation of all protonated carbons within **2** was made possible from the HMQC experiment. The assignment of all quaternary carbons as well as the connectivities of the structural fragments of **2** were secured from HMBC correlations (Table 2). For example, HMBC correlations of H₂-2/C-7, H₂-6/C-7, H₃-8/C-7 (δ 176.41) and H-2'/C-3', H-5'/C-3', H-6'/C-3' (δ 146.78) established the assignments of these signals. Similarly, HMBC cross-peaks of H-2'/C-4', H-5'/C-4', H-6'/C-4' (δ 149.54), and H-7'/C-9', H-8'/C-9' (δ 168.91) confirmed these assignments (Table 2).

The above-mentioned data are in good agreement with those reported for 3-caffeoylquinic acid methyl ester¹⁰ and this is the first report of this compound in the family Amaryllidaceae.

The 2D NOESY experiment of 2 showed cross-peaks of H-3/H-5 and H-4/H-5, supporting the configuration of the quinic acid moiety of 2 (Figure 1).

The existence of 3-caffeoylquinic acid methyl ester in *Pancratium maritimum* is noteworthy from the viewpoint of chemotaxonomy of the genus and family Amaryllidaceae.

Biological evaluation of 1 and 2 1- Antitumor activity

The *in-vitro* cytotoxicity of compounds 1 and 2 against HeLa cells were carried out according to Fukuzawa's method.¹¹

A) Preparation of test samples

Tested samples for screening were prepared as follows. The polar crude extracts were dissolved in a mixture of $H_2O/MeOH$ (1:1) to make a sample of 10 mg/mL. The lipophilic/organic crude extracts were dissolved in EtOH to make a solution of 5 mg/mL. Column fractions and pure compounds were prepared by dissolving in MeOH, DMSO, or water to afford final concentration of 1 mg/mL.

B) Cell culture

Human cervical carcinoma (HeLa) cells were maintained in adhesion on Petri dishes with Minimun Essential Medium (Gibco) supplemented with 10% fetal bovine serum, 2 μ g/mL gentamycin, 2 μ g/mL antibioticantimicotic, and 0.3 M NaHCO₃ (adjusted to pH 7.0-7.4 with 2 M HCl). Subculture was made twice a week to maintain regular proliferation.

C) Cytotoxicity test

HeLa cells in a 200 mL of medium (cell concentration 625 cells/mL) were plated on 96well plates and allow to adhere at 37° under an atmosphere of a mixture of air/CO₂ (95:5) for 24 hours. 2 mL of sample solution (concentration of 1 mg/mL) was added, fold serial dilution was made, and the cells were incubated for 72 h. Positive cytotoxicity control was also prepared using adriamycin (1 mg/mL in DMSO).

The cell's viability was assessed through an MTT conversion method. After 72 h, MTT (1 mg/mL, 50 mL) was added and the cells were incubated for an additional 3-4 h. The mixtures of medium and MTT solutions were removed by aspiration, and diluted with 150 mL DMSO. The optical density of each well was examined with а microplate spectrophotometer at wavelength of 510 nm. The IC_{50} values were estimated using SOFTmax Pro3.1.1 program or plotting % inhibitions at tested concentrations on semi-log graph. Percent inhibition was calculated as $\{(A-B)/A \times 100\}$, where A and B were absorbance of negative control and sample, respectively.

The cytotoxicity of compounds 1 and 2 against HeLa cells was evaluated as IC_{50} values and was presented in Table 3. Adriamycin was used as a positive cytotoxicity control. Pancratistatin (1) and 3-caffeoylquinic acid methyl ester (2) showed potent cytotoxicity against HeLa cells with $IC_{50} = 0.06$ and 1.0 μ g/mL, respectively. The positive cytotoxicity control (Adramycin) showed IC_{50} value = 0.066 μ g/mL.

It is known that pancratistatin possesses a wide range of *in-vitro* and *in-vivo* cancer cell growth inhibitory as well as antiviral activities.^{9,12,13} But, this is the first report of its activity against HeLa cells. In addition, this is

the first report of the cytotoxic activity of 3-caffeoylquinic acid methyl ester.

Table 3:	Cytotoxicity of compounds 1 and 2		
against Hela cells (IC ₅₀ : μ g/mL).			

	Compound 1	Compound 2	Adriamycin*
HeLa Cells	0.06	1.0	0.066

*Positive cytotoxicity control.

2- Anti-Mycobacterium tuberculosis activity

Compounds 1 and 2 were submitted to the Tuberculosis Antimicrobial Acquisition and Coordinating Facility (TAACF) of the Southern Research Institute for biological studies with Mycobacterium tuberculosis $H_{37}Rv.$ In-vitro evaluation of anti-Mycobacterium tuberculosis activity of 1 and 2 was carried out according to Alamar Blue Assay.¹⁴ Below is a brief description of the assay:

A) Alamar blue susceptibility test (MABA)¹⁴

Antimicrobial susceptibility testing was performed in black, clear-bottomed, 96-well microplates (black view plates; Packaed Instrument Company, Meriden, Conn.) in order to minimize background fluorescence. Outer perimeter wells were filled with sterile water to prevent dehydration in experiment wells. Initial drug dilutions were prepared in either DMSO or distilled deionized water, and subsequent twofold dilutions were performed in 0.1 mL of 7H9GC (no Tween 80) in the microplates. BACTEC 12B-passaged inocula were initially diluted 1:2 in 7H9GC, and 0.1 mL was added to wells. Subsequent determination of bacterial titers yielded 1 x 10⁶ CFU/mL in plat wells for H₃₇Rv. Frozen inocula were initially diluted 1:20 in BACTEC 12B medium followed by 1:50 dilution in 7H9GC. Addition of 1/10 mL to wells resulted in final bacterial titers of 2.0 x 10^{5} CFU/mL. Wells containing drug only were used to detect autofluorescence of compounds. Additional control wells consisted of bacteria only (B) and medium only (M). Plates were incubated at 37 °C. Starting at day 4 of incubation, 20 µL of 10 x AlamarBlue solution (Alamar Biosciences/Accumed, Westlake. Ohio) and 12.5 µL of 20% Tween 80 were added to one B well and one M well, and plates were reincubated at 37 °C. Wells were observed at 12 and 24 h for a color change from blue to pink and for a reading of 50,000 fluorescence units (FU). Fluorescence was measured in a Cytofluor II microplate (PerSeptive Biosystems, fluorometer Framingham, Mass.) in bottom-reading mode with excitation at 530 nm and emission at 590 nm. If the B wells became pink by 24 h, reagent was added to the entire plate. If the well remained blue or of 50,000 FU was measured, additional M and B wells were tested daily until a color change occurred, at which time reagents were added to all remaining wells. Plates were then incubated at 37 °C, and results were recorded at 24 h post-reagent addition. Visual MICs were defined as the lowest concentration of drug that prevented a color change. For fluorometric MICs, a background subtraction was performed on all wells with a mean of triplicate M wells. Percent inhibition was defined as {1- (test well FU/mean FU of triplicate B wells) x 100}. The lowest drug concentration effecting an inhibition 90% was considered the MIC.

B) Statistical analysis

All analysis were performed with the program SAS (SAS Institute Inc., Cary, N.C.). Correlation coefficient were defined according to Spearman for ranked data analysis and Pearson for raw data analysis to determine differences between the BACTEC system and the MABA either fluorometrically or visually for determination of MIC.¹⁵ A general linear model procedure using analysis of variance of ranked data was performed for each of the three variables of MIC generation to determine significant differences between the replicate comparisons of the four experiments. Tukey's Studentized range test was also used for pairwise post hoc comparison of variable analysis of ranked measure for significant differences among techniques of MIC determination and for differences between techniques for individual antimicrobial agents for each bacterial strain. Significance was determined at P = 0.05.

C) Bacterial strains and growth conditions

M. tuberculosis $H_{37}Rv$ ATCC 27294 ($H_{37}RV$), was obtained from the American

Type Culture Collection (Rockville, Md.). For the first three (of four) replicate experiments, $H_{37}Rv$ inocula were first passaged in radiometric 7H12 broth (BACTEC 12B; Becton Dickinson Diagnostic Instrument Systems, Sparks, Md.) until the growth index (GI) reached 800 to 999.

For the fourth replicate experiment, $H_{37}Rv$ was grown in 100 mL of Middlebrook 7H9 broth (Difco, Detroit, Mich.) supplemented with 0.2% (vol/vol) glycerol (Sigma Chemical Co., Saint Louis. Mo.), 10% (v/v) OADC (oleic acid, albumin, dextrose, catalase; Difco), and 0.05% (v/v) Tween 80 (Sigma). The complete medium was referred to as 7H9GC-Tween. Cultures were incubated in 500-mL nephelometer flasks on a rotary shaker (New Brunswick Scientific, Edison, N.J.) at 150 rpm and 37 °C until they reached an optical density of 0.4 to 0.5 at 550 nm. Bacteria were washed and suspended in 20 mL of phosphate-buffered saline and passed through an 8-µm-pore-size filter to eliminate clumps. The filtrates were aliquoted, stored at -80 °C, and used within 30 days.

Compound **1** was found to effect moderate inhibitory activity (48%) against *M. tuberculosis* H_{37} Rv at a concentration of 12.5 µg/mL, whereas **2** inhibited 43% of the mycobacterial growth at the same concentration.

This is the first report of anti-tuberculosis activity of an Amaryllidaceae alkaloid (pancratistatin) and for 3-caffeoylquinic acid methyl ester.

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